This is the Author's Pre-print version of the following article: Angel M. Lopez-Hidalgo, Victor E. Balderas Hernandez, Antonio De Leon-Rodriguez, Scale-up of hydrogen and ethanol co-production by an engineered Escherichia coli, Fuel, Volume 300, 2021, 121002, which has been published in final form at: <u>https://doi.org/10.1016/j.fuel.2021.121002</u>

© 2021 This manuscript version is made available under the Creative Commons Attribution-NonCommercial-NoDerivatives 4.0 International (CC BY-NC-ND 4.0) license <u>http://creativeco.mmons.org/licenses/by-nc-nd/4.0/</u>

1	Scale-up of hydrogen and	ethanol co-production by	an engineered Escherichia coli

- 2
- 3

4 Angel Mario Lopez-Hidalgo^{1,a}, Victor E. Balderas Hernandez¹, Antonio De Leon5 Rodriguez^{1*}

6

¹División de Biología Molecular, Instituto Potosino de Investigación Científica y 7 Tecnológica, Camino a la Presa San José 2055, Col. Lomas 4ª sec., C.P. 78216, San Luis 8 Potosí, SLP., México. 9 ^a Present address: Laboratorio de Futuros en Bioenergía, Centro de Investigación y de 10 Estudios Avanzados del Instituto Politécnico Nacional, Unidad Guadalajara de Ingeniería 11 Avanzada, Av. del Bosque 1145, Colonia el Bajío, Zapopan 45019, Jalisco, México 12 13 Submitted to: Fuel. 14 15 16 *Corresponding author. E-mail address: aleonr@me.com, aleonr@ipicyt.edu.mx. 17

- 18 Tel.: +52 444 8342044
- 19

20 Abstract

In this work, the scale-up from 0.01 to 10 L process for the co-production of hydrogen (H₂) 21 and ethanol (EtOH) by a genetically engineered Escherichia coli that utilizes 22 23 hemicellulosic hydrolysates from wheat straw as substrate is presented. Co-production of 24 biofuels was performed through the redirection of carbon-flux to ethanol by deleting *ldhA* (D-lactate dehydrogenase) and frdD (fumarate reductase) genes in an H2-overproducer 25 26 strain (E. coli WDH). Resulting strain, E. coli WDH-LF ($\Delta hycA \Delta frdD \Delta ldhA$), increased up to 70% and 167% the H_2 and EtOH production compared with the parenteral strain. The 27 28 yields of H₂ and EtOH remained constant at all the evaluated scales. In 10 L bioreactors, the production parameters such as maximum production, production rate and yield were 29 30 5603.0 ± 233.5 mL H₂/L, 41.4 ± 4.0 mL H₂/L·h, 342.7 ± 14.3 mL H₂/g TRS, 7.90 ± 0.28 g EtOH/L and 0.48 \pm 0.01 g EtOH/g TRS, respectively. The results demonstrate the potential 31 of the co-production of H_2 and EtOH at different production scales by the engineered E. 32 33 coli strain using lignocellulosic biomass as feedstock, such as wheat straw hydrolysates. 34

35 Keywords: Hemicellulosic hydrolysates, dark fermentation, metabolic engineering,

36 biofuels, phage P1 transduction

38 1 Introduction

The main criteria for the selection of feedstock for biofuels production are their availability. 39 cost, carbohydrate content and biodegradability. Lignocellulosic biomass (LCB) meets with 40 41 them and it is the most abundant renewable resource on earth [1,2]. From them, wheat straw is one of the most abundant LCB worldwide, with a production of 529 million tons 42 [3]. The greater production of agricultural residues is obtained from cereals (e.g., wheat43 44 among others) and is directly associated with the production of grains. For every kilogram of grain produced, approximately one kg of lignocellulosic residue (*i.e.* straw) is obtained 45 [4]. At the end of 2019, Agricultural and Fisheries Information Service (SIAP by its 46 acronym in Spanish) reported a production of 3 million tons of wheat-grain [5]. Their 47 residues are mainly used as fodder in livestock, as well soils improvers, construction 48 materials, and composting [4]. Notwithstanding, with the search for alternative and 49 renewable energy sources, the use of agricultural residues as feedstock for biofuels 50 production is seen as a prominent option. 51

52 For biorefining purposes, the complex LCB lignocellulosic matrix must be debilitated and deconstructed to simpler chemical forms by pre-treatment methods before further 53 processing [6,7]. After pre-treatment, the simple carbohydrates from LCB can be converted 54 55 into biofuels through fermentation. Microbial fermentations offer an attractive alternative to produce sustainable energy because metabolic diversity of microorganisms enables the use 56 of different substrates as the starting point for biofuels (*e.g.*, ethanol, hydrogen, biodiesel, 57 among others) production. However, high yields and productivities in every step during 58 59 biofuel production by microorganisms are required for profitable industrial-scale 60 production. Genetically modified microorganisms with redesigned metabolic pathways can improve both yield and productivity [8–10]. *Escherichia coli* has the capability to generate 61

simultaneously hydrogen (H_2) and ethanol (EtOH), because the anaerobic pyruvate 62 breaking produces formate and acetyl-CoA. Then, H₂ is produced by formate 63 hydrogenlyase system through the decarboxylation of formate [11], while EtOH formation 64 65 is carried out by the action of alcohol dehydrogenase, which catalyzes the reaction from acetyl-CoA to EtOH [12]. However under anaerobic conditions, other end-products such as 66 lactate, acetate, and succinate are synthetized to maintain the NADH/NAD⁺ balance and 67 68 intracellular pH [13]. Since these metabolites are reduced molecules, their synthesis reduce both H_2 and EtOH production [14]. Therefore, the removal of the synthesis of one or more 69 70 of these end-products will increase the carbon flux towards H_2 and EtOH synthesis.

Although metabolic engineering in *E. coli* have been widely studied and employed to improve biofuels performance [14–16], to our knowledge, studies on the scale-up have not been reported. Therefore, the scale-up of the co-production of H₂ and EtOH by an engineered *E. coli* and using wheat straw hemicellulosic hydrolysates as substrate is presented.

76

77 2 Material and methods

78 2.1 Lignocellulosic biomass

Wheat straw (WS) used in this work was harvested in the spring of 2017 in La Barca
(Jalisco, Mexico). The feedstock was milled with a hammer mill (Azteca 301012) using a
1.27 cm screen. LCB composition was determined according to NREL laboratory analytical
procedures [17]. Glucan, xylan and lignin content in the LCB (dry basis) were 41.39, 21.0
and 16% for WS, respectively.

85 2.2 Hydrolysis of hemicellulosic hydrolysate

The hydrolysate was obtained by auto-hydrolysis followed by a 0.25% H₂SO₄ pre-treatment 86 [18]. Auto-hydrolysis was carried out in a semi-pilot scale pre-treatment continuous tubular 87 88 reactor (PCTR) at 150 psi (about 180°C) with a mean residence time of 18 min [18]. The biomass from auto-hydrolysis was further hydrolyzed with 0.25% (v/v) H₂SO₄ in a 1:2 89 (w/v) solids loading ratio, in an autoclave at 121°C for 60 min. The liquid fraction from the 90 91 pre-treatment was separated by centrifugation and collected for further analysis, and it was identified as wheat straw pretreated (WSP); which contained 1.8 g/L glucose, 39.8 g/L 92 xylose, 7.9 g/L arabinose, 2.6 g/L formate, 7.8 g/L acetate and 1.2 g/L furfural. Further 93 dilutions were made to obtain a hydrolysate with a final concentration of 16.1 ± 0.2 g/L of 94 total reducing sugars (TRS) to perform the scale-up experiments. 95

96

97 2.3 Construction of mutant strains

Strains, plasmids, and primers used for the construction of the mutant strains are shown in 98 99 Table 1. W3110 $\Delta frdD$ and W3110 $\Delta ldhA$ strains from Keio collection [19] were used as donor. The deletion of *frdD* and *ldhA* genes was achieved using P1 transduction method 100 [20], with some modification as follows: To prepare liquid P1 lysate, an overnight culture 101 102 of E. coli donor strain (W3110 $\Delta frdD$ or W3110 $\Delta ldhA$) in LB broth with 30 mg/L kanamycin was washed and suspended in MC medium (10 mM MgSO₄ and 5 mM CaCl₂). 103 Phage P1 was added with multiplicity of infection between 0.1 and 1.0, and the culture was 104 incubated for 30 min at room temperature. After incubation, the culture was added into soft 105 106 agar and finally plated in LB agar plates and incubated overnight at 37°C. After lysis, the 107 culture was treated with chloroform and the debris was removed by centrifugation. The phage was stored at 4° C until its application. For transduction, an overnight culture of E. 108

109 *coli* recipient strain (WDH or WDHL) in LB medium was washed and suspended in MC 110 medium. Phage P1 lysate with the donor strain (W3110 $\Delta frdD$ or W3110 $\Delta ldhA$) was added 111 with multiplicity of infection between 0.1 and 1.0. The phage was allowed to absorb for 30 112 min at room temperature, and then 1 M sodium citrate was added. Finally, all the mixture 113 was plated on selective LB plates with 30 mg/L kanamycin and incubated at 37°C until 114 colonies appeared.

115 The deletion of *frdD* and *ldhA* genes was verified by colony PCR with OGF-F, OGF-R, 116 OG-L-F and OG-L-R primers (Table 1). The $\Delta frdD$, $\Delta ldhA$, or $\Delta ldhA \Delta frdD$ strains were 117 transformed by electroporation with the pCP20 plasmid [22], and selected by ampicillin 118 resistance in LB + 30 mg/L ampicillin plates at 28°C. Kanamycin resistance loss was 119 verified by subcultivation on LB plates plus 30 mg/L kanamycin. Selected colonies were 120 tested by PCR to confirm the loss of kanamycin resistance.

121

122 2.4 Effect of *frdD* and *ldhA* genes deletion on co-production of H₂ and EtOH

123 The effect of frdD and ldhA genes deletion on co-production of H₂ and EtOH by Escherichia coli strains (Table 1) was determined by using glucose as substrate (20 g/L). 124 The experiments were done in anaerobic serological bottles containing 0.01 L of B buffer 125 126 [21], 1 mL/L trace elements solution [22], 0.01 g/L MgSO₄ and 1 g/L yeast extract (Difco). The cultures were started with an optical density of 0.2 measured at a wavelength of 600 127 nm (OD₆₀₀), pH 7.5, and they were incubated at 31°C and 175 rpm, until no longer 128 129 production of H₂ was observed. All the experiments were carried out in triplicate. The production of H₂, EtOH, and other metabolites was measured as indicated in section 2.6. 130

131

132 **2.5** Scale-up of H₂ and EtOH co-production

E. coli WDH–LF strain was selected to perform the scale-up of H_2 and EtOH co-production 133 using WSP as substrate (16 g/L TRS). The working volumes used were 0.01, 0.1, 1 and 10 134 135 L. In 0.1, 1 and 10 L experiment, a volume of the headspace relative to that of the liquid approximately of 40% was used and for 0.01 L reactors was 20%. The experiments were 136 137 done under anaerobic conditions using the B buffer [21] plus 1 mL/L trace elements 138 solution [22], 0.01 g/L MgSO₄ and 1 g/L yeast extract (Difco). The cultures were started with an OD_{600} of 0.2 and pH 8.2, incubating them at 31°C and shaking at 200 rpm, until no 139 140 longer production of H₂ was observed. 10 L bioreactors were stirred at 250 rpm during first 29 h and then it was increased at 400 rpm. Batch cultures with working volumes 1 and 10 L 141 142 were performed in 1.5-L and 13.5-L bioreactors equipped with two six-blade Rushton turbines flat, with a H/D ration of 2.1 and 1.5, respectively. The pH was monitored using an 143 autocleavable electrode (Applikon[®] Biotechnology) and connected to Bioconsole ADI 1035 144 (Applikon[®] Biotechnology) controlled by the ADI 1030 Biocontroller (Applikon[®] 145 Biotechnology). BioXpert 1.3 software (Applikon[®] Biotechnology) was used for data 146 acquisition. The experiments in serological bottles and bioreactors were carried out in 147 quadruplicate and duplicates, respectively. The production of H₂ and EtOH was measured 148 149 as indicated in Section 2.6.

- 150
- 151 **2.6 Analytical Methods**

TRS determination was performed by the dinitrosalicylic acid (DNS) method [23], with some modifications, as follow: 250 μ L of diluted sample with 750 μ L of DNS reagent (10 g/L NaOH, 200 g/L KNaC4H4O6·4H2O, 0.5 g/L Na2S2O5, 2 g/L C6H6O, 10 g/L 3,5-Dinitrosalicylic acid) were heated for 5 minutes at 100°C and then cooled down to room temperature. Then, 400 μL of distilled water were added. Xylose (0.1 to 1.0 g/L, Sigma)
was used as the reference standard. The absorbance was measured at 595 nm (iMarkTM
Microplate Absorbance Reader).

159 Simple carbohydrates and metabolites were quantified by an Agilent HPLC equipped with a refractive index (Agilent Technologies 1220 Infinity LC), using a Rezex[™] ROA-Organic 160 161 Acid H+ (Phenomenex) column, operated at 60° C with H₂SO₄ 0.0025 M as a mobile phase 162 (0.550 cm³/min). Furfural was analyzed by Gas Chromatography (Agilent Technologies 6890N Network GC Systems) using a capillary column HP-Innowax ($30 \text{ m} \times 0.25 \text{ mm i.d.}$ 163 164 \times 0.25 m film thickness; Agilent Technologies). Injector and flame ionization detector (FID) temperatures were 220 and 250°C, respectively. Helium was used as carrier gas at 25 165 166 mL/min. Analyses were performed with a split ratio of 10:1 and a temperature program of 35 °C for 2 min, then 10°C/min to 210°C for 1 min. 167

Gas production was measured by acidified water (pH ≤ 2) displacement in an inverted 168 burette connected to serological bottles/bioreactor with rubber tubing and a needle. H₂ 169 170 concentration (%, v/v) in the gas was determined by with a Gas Chromatography (Agilent Technologies 6890N Network GC Systems) coupled to a thermal conductivity detector 171 (Agilent Technologies 6890N Network GC Systems) and using Agilent J&W HP-PLOT 172 173 Molesieve column (0.32 mm ID, 30 m length, 12 µm film) under the following conditions: 200°C, injector temperature; 280°C, detector temperature; 300°C, oven temperature. 174 Helium was used as carrier gas. H₂ volume was corrected to standard conditions of 175 176 temperature and pressure (298.15K and 10^5 Pa).

177

178 2.7 Statistical analysis

For comparisons between samples, the data was analyzed by analysis of variance
(ANOVA) and Tukey's HSD (honestly significant difference) test (Origin[®] 9). A
probability of 5% was accepted as statistically significant.

182

183 **3** Results and discussion

184 3.1 Improvement of H₂ and EtOH co-production by *frdD* and *ldhA* genes deletion in H₂-overproducer *Escherichia coli* strains

186

To enhance the co-production of H₂ and EtOH, the effect of deletion of *ldhA* and *frdD* 187 genes encoding lactate dehydrogenase and fumarate reductase was determined. The 188 modified pathway is shown in Fig. 1A, and the genotypes of the constructed strains are 189 shown in Table 1. The first step it was performed the deletion of *ldhA* gen in two H₂-190 191 overproducer strains, E. coli $\Delta hycA$ (WDH) and E. coli $\Delta hycA$ $\Delta lacl$ (WDHL) [22], through P1 transduction method (Fig. 1B). Then, $\Delta hycA \Delta ldhA$ (WDH-L) and $\Delta hycA \Delta lacI$ 192 $\Delta ldhA$ (WDHL-L) E. coli strains were generated. The effect of ldhA gen deletion on co-193 194 production of H₂ and EtOH was determined in batch cultures (0.1 L) using glucose (20 g/L) 195 as substrate at 37°C and initial pH of 7.5. After 160 h (Fig. 2A), the WDH-L and WDHL-L strains produced 1639.7 \pm 231.3 and 1870.5 \pm 286.0 mL H₂/L, respectively. According to 196 197 the analysis of variance, not statistically significant difference (Table A1, p < 0.05) was found in the H₂ production achieved by each strain. The H₂ production rate (rH₂) obtained 198 by WDHL-L strain was 1.7-fold higher than the achieved by WDH-L strain (Fig. 2A). 199 However, comparing it with their parental (WDH and WDHL) strains [22,24], rH₂ seems to 200 not have been affected by *ldhA* gen deletion, which agrees with previously reported for 201

202 *ldhA E. coli* mutants strains [25,26]. Regarding to metabolites distribution at the end of dark fermentation (Fig. 2B), the WDH-L strain produced 1.20 ± 0.10 g/L succinate, $0.90 \pm$ 203 0.45 g/L lactate, 3.18 \pm 0.33 g/L acetate and 5.90 \pm 0.75 g/L ethanol. Whereas, the WDHL-204 205 L strain achieved 1.58 ± 0.02 g/L succinate, 3.60 ± 0.34 g/L lactate, 3.82 ± 0.13 g/L acetate 206 and 3.60 ± 0.30 g/L ethanol. The end pH was ranging between 4.75 and 4.93. According to 207 the results, the WDH-L strain produced 1.6-fold more ethanol than the WDHL-L strain. 208 There was statistically significant difference in the final concentration of the metabolites in the cultures after 160 h of fermentation (Table A2). Both strains (WDH-L and WDHL-L) 209 210 achieved approximately the same cumulative H₂ production, but the WDH-L strain produced the higher concentration of EtOH. This finding helps us to select WDH strain for 211 212 the following experiments.

The deletion of *frdD* gen was performed in WDH and WDH-L strains to obtain the WDH-F 213 (E. coli $\Delta hycA \Delta frdD$) and WDH-LF (E. coli $\Delta hycA \Delta ldhA \Delta frdD$) strains. To determine 214 the effect of *frdD* gen deletion on co-production of H₂ and EtOH by *E. coli*, the 215 216 fermentation of 20 g/L of glucose in a batch culture during 180 h at 31°C and initial pH of 7.5 was conducted (Fig. 3). The WDH-F strain obtained 1.3- and 1.4-fold more H₂ than the 217 wild-type (E. coli W3110, WT) strain and its parental (WDH) strain, respectively. 218 219 Regarding to rH₂, the WT, WDH, and WDH-F obtained almost the same results –ranging between 53.9 and 59.4 mL H₂/L/h without statistical difference, as shows Table A3. 220 However, WDH-F achieved only 50% and 70% of the EtOH obtained by WT and WDH 221 strains, respectively, since flux of glucose was utilized mainly to produce lactate instead 222 EtOH or another metabolite. Comparing both double mutant strains, WDH-F strain 223 224 produced 1.5-fold more H₂ and achieved 2.4-fold higher rH₂ than WDH-L. But the latter

produced 4.5-fold more EtOH than WDH-F due to the fact flux of glucose seems to bepartially redirected to EtOH by the *ldhA* gene deletion.

227 The co-production of H_2 and EtOH was significantly improved by both *ldhA* and *frdD* 228 genes deletion. The WDH-LF strain produced 2,950.3 \pm 261.8 mL H₂/L, 1.7-fold and 1.5-229 fold more hydrogen than the WT and WDH strains, respectively (Fig. 3A). Also, WDH-LF 230 strain produced more H₂ than single mutant strains –WDH-L and WDH-F–, which 231 achieved 1,650.83 \pm 234.20 and 2,523.03 \pm 93.54 ml H₂/L, respectively. Although, H₂ 232 production was increased by ldhA and frdD genes deletion, rH_2 seems to be decreased in 233 WDH-LF strain. WDH-LF strain achieved 40.35 \pm 3.89 mL H₂/L/h, whereas WT and WDH obtained $1.3-53.91 \pm 3.15$ and 55.69 ± 3.23 mL H₂/L/h (Fig. 3B), respectively. 234 235 Succinate and lactate production were dramatically reduced by the deletion of *frdD* and *ldhA* genes, whereas the EtOH production was improved compared to WT strain (Fig. 3C). 236 The WDH-LF strain produced 7.20 ± 0.26 g EtOH/L, which is 2.7-, 3.8-, 1.2, and 5.5-fold 237 more EtOH than obtained by WT, WDH, WDH-L, and WDH-F strains (Table A4), 238 239 respectively. These results are consistent with the previously reported [26,27], and 240 confirmed that the absence of frdD and ldhA genes improves the co-production of H₂ and 241 EtOH by E. coli using glucose as substrate.

Mutagenesis of competing metabolic pathways has been widely employed to improve H₂ and/or EtOH production performance [28]. To improve H₂ production, the most utilized strategies are related to disruption of genes –such as *ldhA*, *frdAB*, *hycA*, *hya*, and *hyb*–, inactivation of formate-hydrogen lyase (FHL) repressor (encoded by *hycA*), and/or overexpression of FHL (encoded by *fhlA*). Regarding to the EtOH production, EtOHproducer strains [*e.g. E. coli* SE2378 ($\Delta ldh \Delta pfl$) and *E. coli* SZ420 ($\Delta frdBC \Delta ldh \Delta ackA$ $\Delta folA-pfl \Delta pdhR::pflBp6-aceEF-lpd$)] have been constructed by deleting the competing

249 fermentation pathways, and implementing another molecular strategies such gene 250 overexpression. However, the low H₂ yields still a limitation for large-scale H₂ production via dark fermentation and it is not sustainable economically. To overcome this issue, co-251 252 production of H₂ and EtOH has been proposed [29]. In this regard, using E. coli $\Delta hycA$ $\Delta hyaAB \Delta hybBC \Delta ldhA \Delta frdAB$ as parental strain, mutant strains devoid *pta-ackA* 253 (encoding to phosphate acetyltransferase and acetate kinase, respectively) or *pfkA* 254 255 (encoding to phosphofructokinase 1) genes were constructed to increase the H₂ and EtOH 256 co-production [30]. The authors concluded that $\Delta ack-pta$ strain does not improve the 257 biofuels co-production. However, the $\Delta pfkA$ strain resulted in an increase of approximately 50% and 9.5% of H₂ and EtOH, respectively. In the present study, we used the *E. coli* Keio 258 259 collection to introduce two mutations into a single strain by the repetition of resistancegene elimination and P1 transduction; this method may be used to engineer E. coli for many 260 applications where multiple chromosomal genes must be eliminated. The deletion of the 261 262 *lhdA* and *frdD* genes in WDH strain, produce an improvement of 70 and 167% on H_2 and 263 EtOH production, respectively.

264

3.2 Scale-up of H₂ and EtOH co-production from wheat straw hemicellulosic

266 hydrolysate

The scaling-up is a vital tool to development bioprocesses since it can reduce errors in the industrial-scale designs, as well as lack of information. For this reason, the effect of the reactor size on co-production of H_2 and EtOH by *E. coli* WDH-LF was determined. Since lignocellulose is a sustainable and worldwide available biomass, it was selected as feedstock to perform the scaling-up of dark fermentation. Therefore, wheat straw hemicellulosic hydrolysate (WSP) obtained by autohydrolysis followed by a 0.25% diluted 273 sulphuric acid pre-treatment was used as substrate for H_2 and EtOH co-production by E. *coli* WDH-LF. The reactor sizes used to scale-up the process were 0.01, 0.1, 1 and 10 L. 274 275 The average of the maximum H₂ concentration (considering all working volumes) was 50.9 276 \pm 7.4 %. The H₂ production obtained in 0.01, 0.1, 1 and 10 L reactors was 5,259.9 \pm 540.2, $4,076.1 \pm 468.4, 5,574.8 \pm 156.6, \text{ and } 5,603.0 \pm 233.5 \text{ mL H}_2/\text{L}$, respectively, as shows in 277 Fig. 4A. H₂ production was similar in all working volumes except in 0.1 L (Table A5), 278 279 which was approximately 25% lower than the other sizes. Regarding to the H₂ yield, the results obtained were 328.7 \pm 33.8, 253.2 \pm 29.1, 350.6 \pm 9.9, and 342.7 \pm 14.3 mL H/g 280 281 TRS for 0.01, 0.1, 1 and 10 L reactors, respectively. H₂ yield seems to be lower in 0.1 L 282 reactors compare with the other reactor sizes (Fig. 4B). These behaviors respond to the fact that the volume of the headspace relative to that of the liquid (working volume) may affect 283 dark fermentation [38,39], because both H₂ and CO₂ accumulate to high pressures in the 284 headspace and inhibits H₂ production [40]. As described in Section 2.5, 0.1 L reactors 285 286 employed lower volume of the headspace relative to that of the working volume than the 287 other reactor sizes. The maximum rH₂ achieved by WDH-LF strain were 40.7 \pm 2.3, 39.2 \pm 5.8, 36.9 ± 1.4 and 41.5 ± 4.0 mL H₂/L·h for 0.01, 0.1, 1 and 10 L reactors, respectively 288 (Fig. 4B). Kinetic of H₂ production are shown in Fig. 5. As noted, 10 L reactor attained the 289 290 longest lag phase compare to the other reactor sizes. This can be explained by the agitation rate utilized in 10 L reactors, which was maintained at 250 rpm during first 29 h and then it 291 was increased at 400 rpm, as described Section 2.5. Mixing may promote dark fermentation 292 293 performance by enhancing liquid-gas mass transfer, heat transfer, as well as 294 homogenization. However, inappropriate stirring can be harmful to H_2 producing bacteria, and thus H_2 production may be inhibited [41,42]. The EtOH productions obtained by 295 WDH-LF strain were 8.9 ± 1.3 , 7.8 ± 1.7 , 7.9 ± 0.6 , and 7.9 ± 0.3 g EtOH/L in 0.01, 0.1, 1, 296

and 10 L reactors, respectively. And the EtOH yields achieved during dark fermentation by WDH-LF strain were 0.56 ± 0.08 , 0.48 ± 0.11 , 0.50 ± 0.04 , and 0.48 ± 0.01 in 0.01, 0.1, 1, and 10 L reactors, respectively. The production and yield of EtOH seems not to be affected by the change of reactor size, as shown in Fig. 4C.

The pH profile of H₂ and EtOH co-production for 1 and 10 L appear in Fig. 6. In 1 L 301 302 reactors, H₂ and EtOH production started in the first 15 h of fermentation, while for 10 L 303 reactors, the first evidence of H₂ and EtOH production was after 36 h of fermentation. The pH decreased only two units due to the low organic acids production, which is a 304 305 consequence of the *frdD* and *ldhA* genes absence, involved in the succinate and lactate production, respectively. The optimal conditions of temperature and initial pH used in this 306 307 work were established previously using the WDHL strain [43]. By comparing the results 308 obtained in this work by WDH-LF strain with those previously reported for WDHL strain, the following is noticed: (a) the WDH-LF strain increased 38% the EtOH yield; (b) the H₂ 309 production rate by E. coli WDH-LF was approximately 4-fold than the achieved by E. coli 310 311 WDHL; (c) the H₂ yield increased 2-fold by the deletion of ldhA and frd genes.

312 H₂ yields reported for microbial consortia and pure cultures from lignocellulosic sources, under mesophilic or thermophilic conditions, are in the range of 200 to 500 mL H_2/g 313 314 consumed sugar [44]. Also, theoretical yields between 70 and 99% has been reported for lignocellulosic ethanol achieved by yeast and bacteria [45]. Nonetheless, mainly these 315 biofuels are obtained in separate processes or coupled to the production of other 316 317 components (Table 2). In the present work, it was obtained in one step hydrogen and 318 ethanol with yields up to 342.7 mL H₂/g TRS and 97% (0.48 \pm 0.01 g EtOH/g TRS) of the 319 maximum theoretical of ethanol; this strategy could be used in the conceptual design of 2G biorefineries [29]. Dark fermentation has been shown as the most realistic opportunity to 320

321 leave laboratory scale production behind, which is attributed to the relatively low energy requirements, high biofuels production rates and the chance of using a broad spectrum of 322 organic matter as substrate [46]. However, they still have certain limitations that could be 323 324 crucial for scaling-up, as the stability of the process in self [47,48]. As an answer to the 325 limitations of the dark fermentation, several researchers have proposed the use of 326 genetically modified microorganisms with advanced selected properties [49]. In the present 327 work, we demonstrate that: (1) the increase of biofuels production yields by deletion of the ldhA and frd genes in a H₂-overproducer E. coli strain; (2) E. coli WDH-LF can produce 328 329 efficiently and simultaneously H₂ and EtOH from hemicellulosic hydrolysates via dark fermentation; (3) the yields achieved by WDH-LF strain are stable despite the change in the 330 reactor size. 331

332

333 4 Conclusions

334 Deletion of frdD and ldhA genes positively impact the co-production of H₂ and EtOH by E. 335 coli, since production, production rate and yield of both biofuels were improved. The engineered WDH-LF strain may utilize hemicellulosic hydrolysates as substrates to 336 produce simultaneously and efficiently H_2 and EtOH, under optimal temperature and pH 337 338 conditions. The change in the reactor size have not impact on the co-production of H₂ and EtOH by E. coli WDH-LF using wheat straw as feedstock, which is inferred because the H₂ 339 and EtOH yields were similar despite the working volumes employed. Still, further genetic 340 modifications, as well as engineering techniques must be employed to improve H_2 and 341 342 EtOH production rates and makes them competitive in the actual biofuels framework.

343

344 Acknowledgements

- Partial Financial support is kindly acknowledged to CONACyT-Basicas grant 281700 and
- 346 PDCPN 2014-01-247498. A. Lopez-Hidalgo thanks the National Council of Science and
- 347 Technology (CONACYT) for his postdoctoral fellowship 385574. The authors also thanks
- to Arturo Sanchez and Felicia Rodriguez from Bioenergy Futures Laboratory of the Center
- 349 for Research and Advanced Studies of the National Polytechnic Institute, Guadalajara Unit,
- 350 by kindly supply the wheat straw used in this work.

351 **References**

- 352 [1] Cherubini F. The biorefinery concept : Using biomass instead of oil for producing
- energy and chemicals. Energy Convers Manag 2010;51:1412–21.
- doi:10.1016/j.enconman.2010.01.015.
- 355 [2] Kumar R, Singh S, Singh O V. Bioconversion of lignocellulosic biomass:
- Biochemical and molecular perspectives. J Ind Microbiol Biotechnol 2008;35:377–
 91. doi:10.1007/s10295-008-0327-8.
- 358 [3] Kapoor M, Panwar D, Kaira GS. Bioprocesses for Enzyme Production Using Agro-
- 359 Industrial Wastes: Technical Challenges and Commercialization Potential. In:
- 360 Dhillon GS, Kaur S, editors. Agro-Industrial Wastes as Feed. Enzym. Prod. Apply
- 361 Exploit Emerg. Valuab. Use Options Waste Biomass, Elsevier Inc.; 2016, p. 61–93.
 362 doi:10.1016/B978-0-12-802392-1.00003-4.
- 363 [4] Borja-Bravo M, Reyes-Muro L, Espinosa-Garcia JA, Vélez-Izquierdo A. Producción
- 364 y consumo de rastrojos en México. In: Reyes-Muro L, Camacho-Villa TC, Guevara-
- 365 Hernández F, editors. Rastrojos manejo, uso y Merc. en el Cent. y sur México. First
- edit, Ciudad de México: Instituto Nacional de Investigaciones Forestales, Agrícolas
 y Pecuarias; 2013, p. 256.
- 368 [5] Servicio de Información Agroalimentaria y Pesquera. Avance de Siembras y

369 Cosechas. Resumen nacional por cultivo. Ciudad de Mexico: 2019.

- 370 [6] Sivagurunathan P, Kumar G, Mudhoo A, Rene ER, Saratale GD, Kobayashi T, et al.
- Fermentative hydrogen production using lignocellulose biomass: An overview of
- 372 pre-treatment methods, inhibitor effects and detoxification experiences. Renew
- Sustain Energy Rev 2017;77:28–42. doi:10.1016/j.rser.2017.03.091.
- [7] Kumar G, Shobana S, Nagarajan D, Lee DJ, Lee KS, Lin CY, et al. Biomass based

375		hydrogen production by dark fermentation — recent trends and opportunities for
376		greener processes. Curr Opin Biotechnol 2018;50:136-45.
377		doi:10.1016/j.copbio.2017.12.024.
378	[8]	Liao JC, Mi L, Pontrelli S, Luo S. Fuelling the future: Microbial engineering for the
379		production of sustainable biofuels. Nat Rev Microbiol 2016;14:288-304.
380		doi:10.1038/nrmicro.2016.32.
381	[9]	Shaw AJ, Lam FH, Hamilton M, Consiglio A, MacEwen K, Brevnova EE, et al.
382		Metabolic engineering of microbial competitive advantage for industrial
383		fermentation processes. Science (80-) 2016;353:583-6.
384		doi:10.1126/science.aaf6159.
385	[10]	Koskinen PEP, Beck SR, Örlygsson J, Puhakka JA. Ethanol and hydrogen
386		production by two thermophilic, anaerobic bacteria isolated from Icelandic
387		geothermal areas. Biotechnol Bioeng 2008;101:679-90. doi:10.1002/bit.21942.
388	[11]	Balderas-Hernandez VE, Landeros Maldonado KP, Sánchez A, Smoliński A, De
389		Leon Rodriguez A. Improvement of hydrogen production by metabolic engineering
390		of Escherichia coli: Modification on both the PTS system and central carbon
391		metabolism. Int J Hydrogen Energy 2019:1-10. doi:10.1016/j.ijhydene.2019.01.162.
392	[12]	Förster AH, Gescher J. Metabolic engineering of Escherichia coli for production of
393		mixed-acid fermentation end products. Front Bioeng Biotechnol 2014;2:1-12.
394		doi:10.3389/fbioe.2014.00016.
395	[13]	Maeda T, Sanchez-Torres V, Wood TK. Hydrogen production by recombinant
396		Escherichia coli strains. Microb Biotechnol 2012;5:214–25. doi:10.1111/j.1751-
397		7915.2011.00282.x.
398	[14]	Valle A, Cantero D, Bolívar J. Metabolic engineering for the optimization of

- 399 hydrogen production in *Escherichia coli*: A review. Biotechnol Adv 2019;37:616–
- 400 33. doi:10.1016/j.biotechadv.2019.03.006.
- 401 [15] Balderas-Hernandez VE, Landeros Maldonado KP, Sánchez A, Smoliński A, De
- 402 Leon Rodriguez A. Improvement of hydrogen production by metabolic engineering
- 403 of *Escherichia coli*: Modification on both the PTS system and central carbon
- 404 metabolism. Int J Hydrogen Energy 2020;45:5687–96.
- 405 doi:https://doi.org/10.1016/j.ijhydene.2019.01.162.
- 406 [16] Liang L, Liu R, Freed EF, Eckert CA. Synthetic Biology and Metabolic Engineering
- 407 Employing *Escherichia coli* for C2–C6 Bioalcohol Production. Front Bioeng
- 408 Biotechnol 2020;8:710. doi:10.3389/fbioe.2020.00710.
- [17] Sluiter A, Hames B, Ruiz RO, Scarlata C, Sluiter J, Templeton D. Determination of
 Structural Carbohydrates and Lignin in Biomass: Laboratory Analytical Procedure.
- 411 2011. doi:NREL/TP-510-42618.
- 412 [18] Rodríguez F, Amaya-Delgado L, Sanchez A. Xyloligosacharides production from
- 413 lignocellulosic biomass using a pretreatment continuous tubular reactor. Modelling
- and experimental validation. Ind Crop Prod 2019;134:62–70.
- 415 doi:10.1016/j.indcrop.2019.03.058.
- 416 [19] Baba T, Ara T, Hasegawa M, Takai Y, Okumura Y, Baba M, et al. Construction of
- *Escherichia coli* K-12 in-frame, single-gene knockout mutants: The Keio collection.
 Mol Syst Biol 2006;2:2006.0008. doi:10.1038/msb4100050.
- 419 [20] Thomason LC, Costantino N, Court DL. E. coli Genome Manipulation by P1
- 420 Transduction. Curr. Protoc. Mol. Biol., 2007, p. 1.17.1–1.17.8.
- 421 doi:10.1002/0471142727.mb0117s79.
- 422 [21] Davila-Vazquez G, de León-Rodríguez A, Alatriste-Mondragón F, Razo-Flores E.

- 423 The buffer composition impacts the hydrogen production and the microbial
- 424 community composition in non-axenic cultures. Biomass and Bioenergy
- 425 2011;35:3174–81. doi:10.1016/j.biombioe.2011.04.046.
- 426 [22] Rosales-Colunga LM, Razo-Flores E, Ordoñez LG, Alatriste-Mondragón F, De
- 427 León-Rodríguez A. Hydrogen production by *Escherichia coli* ΔhycA ΔlacI using
- 428 cheese whey as substrate. Int J Hydrogen Energy 2010;35:491–9.
- 429 doi:10.1016/j.ijhydene.2009.10.097.
- 430 [23] Miller GL. Use of Dinitrosalicylic Acid Reagent for Determination of Reducing
- 431 Sugar. Anal Chem 1959;31:426–8. doi:10.1021/ac60147a030.
- 432 [24] Rosales-Colunga LM, Razo-Flores E, De León Rodríguez A. Fermentation of lactose
- and its constituent sugars by *Escherichia coli* WDHL: Impact on hydrogen
- 434 production. Bioresour Technol 2012;111:180–4. doi:10.1016/j.biortech.2012.01.175.
- 435 [25] Sode K, Watanabe M, Makimoto H, Tomiyama M. Construction and
- 436 Characterization of Fermentative Lactate Dehydrogenase *Escherichia coli* Mutant
- 437 and Its Potential for Bacterial Hydrogen Production. Appl Biochem Biotechnol

438 1999;77:317–24. doi:10.1385/ABAB:77:1-3:317.

- 439 [26] Yoshida A, Nishimura T, Kawaguchi H, Inui M, Yukawa H. Enhanced hydrogen
- 440 production from glucose using ldh- and frd-inactivated *Escherichia coli* strains. Appl

441 Microbiol Biotechnol 2006;73:67–72. doi:10.1007/s00253-006-0456-9.

- 442 [27] Maeda T, Sanchez-Torres V, Wood TK. Enhanced hydrogen production from
- glucose by metabolically engineered *Escherichia coli*. Appl Microbiol Biotechnol
- 444 2007;77:879–90. doi:10.1007/s00253-007-1217-0.
- 445 [28] Majidian P, Tabatabaei M, Zeinolabedini M, Naghshbandi MP, Chisti Y. Metabolic
- 446 engineering of microorganisms for biofuel production. Renew Sustain Energy Rev

447 2018;82:3863–85. doi:10.1016/j.rser.2017.10.085.

- 448 [29] Lopez-Hidalgo AM, Magaña G, Rodriguez F, De Leon-Rodriguez A, Sanchez A.
- 449 Co-production of ethanol-hydrogen by genetically engineered *Escherichia coli* in
- 450 sustainable biorefineries for lignocellulosic ethanol production. Chem Eng J
- 451 2021;406:126829. doi:10.1016/j.cej.2020.126829.
- [30] Seol E, Ainala SK, Sekar BS, Park S. Metabolic engineering of *Escherichia coli*strains for co-production of hydrogen and ethanol from glucose. Int J Hydrogen
 Energy 2014;39:19323–30. doi:10.1016/j.ijhydene.2014.06.054.
- 455 [31] Łukajtis R, Hołowacz I, Kucharska K, Glinka M, Rybarczyk P, Przyjazny A, et al.
- 456 Hydrogen production from biomass using dark fermentation. Renew Sustain Energy
 457 Rev 2018;91:665–94. doi:10.1016/j.rser.2018.04.043.
- 458 [32] Sivagurunathan P, Kuppam C, Mudhoo A, Saratale GD, Kadier A, Zhen G, et al. A
 459 comprehensive review on two-stage integrative schemes for the valorization of dark
 460 fermentative effluents. Crit Rev Biotechnol 2018;38:868–82.
- 461 doi:10.1080/07388551.2017.1416578.
- 462 [33] Soo CS, Yap WS, Hon WM, Ramli N, Md Shah UK, Phang LY. Co-production of
- 463 hydrogen and ethanol by *Escherichia coli* SS1 and its recombinant. Electron J

464 Biotechnol 2017;30:64–70. doi:10.1016/j.ejbt.2017.09.002.

- 465 [34] Sundara Sekar B, Seol E, Park S. Co-production of hydrogen and ethanol from
- 466 glucose in *Escherichia coli* by activation of pentose-phosphate pathway through
- deletion of phosphoglucose isomerase (pgi) and overexpression of glucose-6-
- 468 phosphate dehydrogenase (zwf) and 6-phosphogluconate dehyd. Biotechnol Biofuels
- 469 2017;10:1–12. doi:10.1186/s13068-017-0768-2.
- 470 [35] Seol E, Sekar BS, Raj SM, Park S. Co-production of hydrogen and ethanol from

- 471 glucose by modification of glycolytic pathways in *Escherichia coli* from Embden-
- 472 Meyerhof-Parnas pathway to pentose phosphate pathway. Biotechnol J
- 473 2016;11:249–56. doi:10.1002/biot.201400829.
- 474 [36] Soo C-S, Yap W-S, Hon W-M, Ramli N, Md Shah UK, Phang L-Y. Improvement of
- 475 hydrogen yield of ethanol-producing *Escherichia coli* recombinants in acidic
- 476 conditions. Electron J Biotechnol 2017;26:27–32. doi:10.1016/j.ejbt.2016.12.007.
- 477 [37] Maeda T, Tran KT, Yamasaki R, Wood TK. Current state and perspectives in
- 478 hydrogen production by *Escherichia coli*: roles of hydrogenases in glucose or
- 479 glycerol metabolism. Appl Microbiol Biotechnol 2018;102:2041–50.
- 480 doi:10.1007/s00253-018-8752-8.
- 481 [38] Oh S-E, Zuo Y, Zhang H, Guiltinan MJ, Logan BE, Regan JM. Hydrogen production
- 482 by Clostridium acetobutylicum ATCC 824 and megaplasmid-deficient mutant M5

483 evaluated using a large headspace volume technique. Int J Hydrogen Energy

484 2009;34:9347–53. doi:10.1016/j.ijhydene.2009.09.084.

- 485 [39] Sekoai PT, Yoro KO, Daramola MO. Effect of nitrogen gas sparging on dark
- 486 fermentative biohydrogen production using suspended and immobilized cells of

487 anaerobic mixed bacteria from potato waste. Biofuels 2018;9: 595–604.

- 488 doi:10.1080/17597269.2018.1432275.
- 489 [40] Nemestóthy N, Bélafi-Bakó K, Bakonyi P. Enhancement of dark fermentative H2
- 490 production by gas separation membranes: A review. Bioresour Technol
- 491 2020;302:122828. doi:10.1016/j.biortech.2020.122828.
- 492 [41] Chezeau B, Fontaine JP, Vial C. Analysis of liquid-to-gas mass transfer, mixing and
- 493 hydrogen production in dark fermentation process. Chem Eng J 2019;372:715–27.
- doi:10.1016/j.cej.2019.04.191.

- [42] Chezeau B, Vial C. Combined effects of digestate viscosity and agitation conditions
 on the fermentative biohydrogen production. Biochem Eng J 2019;142:105–16.
- 497 doi:10.1016/j.bej.2018.11.016.
- 498 [43] Lopez-Hidalgo AM, Sánchez A, De León-Rodríguez A. Simultaneous production of
- 499 bioethanol and biohydrogen by *Escherichia coli* WDHL using wheat straw
- 500 hydrolysate as substrate. Fuel 2017;188:19–27. doi:10.1016/j.fuel.2016.10.022.
- 501 [44] Cheng CL, Lo YC, Lee KS, Lee DJ, Lin CY, Chang JS. Biohydrogen production
- from lignocellulosic feedstock. Bioresour Technol 2011;102:8514–23.
- 503 doi:10.1016/j.biortech.2011.04.059.
- 504 [45] Paulova L, Patakova P, Branska B, Rychtera M, Melzoch K. Lignocellulosic
- ethanol: Technology design and its impact on process efficiency. Biotechnol Adv
 2015;33:1091–107. doi:10.1016/j.biotechadv.2014.12.002.
- 507 [46] Sivagurunathan P, Kumar G, Bakonyi P, Kim SH, Kobayashi T, Xu KQ, et al. A
- 508 critical review on issues and overcoming strategies for the enhancement of dark
- 509 fermentative hydrogen production in continuous systems. Int J Hydrogen Energy

510 2016;41:3820–36. doi:10.1016/j.ijhydene.2015.12.081.

- 511 [47] Jarunglumlert T, Prommuak C, Putmai N, Pavasant P. Scaling-up bio-hydrogen
 512 production from food waste: Feasibilities and challenges. Int J Hydrogen Energy
- 513 2018;43:634–48. doi:10.1016/j.ijhydene.2017.10.013.
- 514 [48] Tapia-Venegas E, Ramirez-Morales JE, Silva-Illanes F, Toledo-Alarcón J, Paillet F,
- 515 Escudie R, et al. Biohydrogen production by dark fermentation: scaling-up and
- 516 technologies integration for a sustainable system. Rev Environ Sci Biotechnol
- 517 2015;14:761–85. doi:10.1007/s11157-015-9383-5.
- 518 [49] Ntaikou I, Antonopoulou G, Lyberatos G. Biohydrogen production from biomass

and wastes via dark fermentation: A review. Waste and Biomass Valorization 519 2010;1:21-39. doi:10.1007/s12649-009-9001-2. 520 521 Maeda T, Sanchez-Torres V, Wood TK. Metabolic engineering to enhance bacterial [50] 522 hydrogen production. Microb Biotechnol 2008;1:30-9. doi:10.1111/j.1751-523 7915.2007.00003.x. 524 Romero Soto L, Byrne E, van Niel EWJ, Sayed M, Villanueva CC, Hatti-Kaul R. [51] 525 Hydrogen and polyhydroxybutyrate production from wheat straw hydrolysate using 526 Caldicellulosiruptor species and Ralstonia eutropha in a coupled process. Bioresour 527 Technol 2018;272:259-66. doi:10.1016/j.biortech.2018.09.142. Valdez-Vazquez I, Pérez-Rangel M, Tapia A, Buitrón G, Molina C, Hernández G, et 528 [52] al. Hydrogen and butanol production from native wheat straw by synthetic microbial 529 530 consortia integrated by species of Enterococcus and Clostridium. Fuel 2015;159:214-22. doi:10.1016/j.fuel.2015.06.052. 531 532 Toquero C, Bolado S. Effect of four pretreatments on enzymatic hydrolysis and [53] 533 ethanol fermentation of wheat straw. Influence of inhibitors and washing. Bioresour Technol 2014;157:68–76. doi:10.1016/j.biortech.2014.01.090. 534 Pawar SS, Nkemka VN, Zeidan AA, Murto M, Van Niel EWJ. Biohydrogen 535 [54] 536 production from wheat straw hydrolysate using Caldicellulosiruptor saccharolyticus 537 followed by biogas production in a two-step uncoupled process. Int J Hydrogen Energy 2013;38:9121-30. doi:10.1016/j.ijhydene.2013.05.075. 538 539

540 **Figure captions**

Fig. 1. Carbon flux in *E. coli* during anaerobic metabolism and the strategy used to 541 improve the co-production of H₂ and EtOH. (A) Improving the co-production of H₂ and 542 EtOH by deletion of *ldhA* and *frdD* genes. The *dashed orange lines* are the pathways 543 inactivated, and the bold yellow lines are the pathways enhanced. Chemical structures are 544 shown for dark fermentation products and pyruvic acid. (B) Escherichia coli genome 545 manipulation by phage P1 transduction. Homologous regions are indicated by red and 546 green boxes; dtg, deletion target gen (*ldhA* and *frdD* genes for this study); res, antibiotic 547 resistance gen (kanamycin for this study). Flp recognition sites are indicated. 548

549

Fig. 2. Effect of *ldhA* gene deletion on the H₂ and metabolites production by WDH and WDHL strains. Kinetic of H₂ production (A) and final concentration of metabolites (B). Data are presented as mean \pm standard deviation. Batch culture were done in 0.1 L anaerobic serological bottles using glucose (20 g/L) as substrate at 37°C and initial pH of 7.5.

555

Fig. 3. Improvement of H₂ and EtOH co-production by metabolic engineering in *E. coli* strains. Kinetic of H₂ production (D), cumulative H₂ production (E) and final concentration of metabolites (F). Data are presented as mean \pm standard deviation. * indicates statistically significant differences compared with the wild type (WT) cultures (*p* < 0.05). Batch culture were done in 0.1 L anaerobic serological bottles using glucose (20 g/L) as substrate at 37°C and initial pH of 7.5.

562

Fig. 4. Effect of the increase of reactor size on co-production of H₂ and EtOH by *E. coli* WDH-LF using WSP as substrate. Batch cultures were done in 0.01, 0.1, 1 and 10 L
at 31°C and initial pH of 8.2. Production (A), production rate and yield (B) of H₂.
Production and yield of EtOH (C). Data are presented as mean ± standard deviation.

Fig. 5. Kinetics of H₂ production by *E. coli* WDH-LF using WSP hydrolysate as
substrate. Batch cultures were done in 0.01, 0.1, 1 and 10 L at 31°C and initial pH of 8.2.

Fig. 6. pH profile during co-production of H₂ and EtOH by *E. coli* WDH-LF using
WSP hydrolysate as substrate. Batch cultures were done in 1 and 10 L at 31°C.

Table 1 573

Strains	Relevant genotype	Source	
WT	Escherichia coli W3110 (lac ⁺ , gal ⁺ , F ⁻ , ⁻ IN (rrnD-	Laboratory stock	
	rrnE)1, rph-1)		
W3110 ∆ <i>frdD</i>	WT $\Delta fr dD$	Laboratory stock	
W3110 $\Delta ldhA$	WT $\Delta ldhA$	Laboratory stock	
WDH	WT $\Delta hycA$	[22]	
WDHL	WT $\Delta hycA \Delta lacI$	[22]	
WDHL-L	WT $\Delta hycA \Delta lacI \Delta ldhA$	This work	
WDH–L	WT $\Delta hycA \Delta ldhA$	This work	
WDH–F	WT $\Delta hycA \Delta frdD$	This work	
WDH-LF	WT $\Delta hycA \Delta ldhA \Delta frdD$	This work	
Plasmid			
pCP20	FLP recombinase expression plasmid (bla, cat)	[50]	
	$p_{\rm R}$ FLP ⁺ , cI857 ⁺ , pSC101 ori TS		
Primers	Sequence		
OGF-F	GAGGGGCAGCAAATGTGGAG	This work	

TGAACTGGCACCGAAAGCGG

CGCGGCTACTTTCTTCATTG

GGTTGCGCCTACACTAAGCAT

This work

This work

This work

57

575

OGF-R

OG-L-F

OG-L-R

577 **Table 2**

578 Comparison of the production and yield of hydrogen and ethanol from wheat straw biomass.

Inoculum	Substrate treatment	Operation mode	Working volume (L)	Hydrogen production (mL H ₂ /L)	Hydrogen yield (mL H ₂ /g consumed substrate)	Ethanol production (g EtOH/L)	Ethanol yield (g EtOH/g consumed sugars)	Reference
Caldicellulosiruptor saccharolyticus and C. owensensis	Steam explosion (190 °C, 10 min), enzymatic hydrolysis	Continuous pH 6.5, 70°C	1.5	3350 ^a (134 mmol H ₂ /L)	-	-	-	[51]
A three-species culture of epiphytic <i>Enterococcus</i>	Hydration of fibres	Batch pH 6, 37°C	0.1	-	386	-	-	[52]
<i>Clostridium beijerinckii</i> and <i>C. cellulovorans</i>	Biologically treated WS enriched in cellulose	Batch pH 5.5, 37°C	0.1	-	-	3.7	-	
Pichia stipites	5% H ₂ O ₂ (pH 11.5, 50°C, 120 rpm, 60 min)	Batch pH 5, 30°C	0.025	-	-	17.37	0.44	[53]
Caldicellulosiruptor saccharolyticus	Acid pretreatment (H ₃ PO ₄ , 190°C, 5 min), enzymatic hydrolysis	Continuous 6.5 pH, 70°C	1	-	413 ^a (3.04 mol H ₂ /mol sugars)	-	0.05 ^a (0.19 mol EtOH/mol sugars)	[54]
Escherichia coli WDH-LF	Autohydrolysis (180°C, 150 psi, retention time: 18 min) followed by acid hydrolysis	Batch pH 8.2,	0.01 0.1	$5259.9 \pm 540.2 \\ 4076.1 \pm 468.4$	328.7 ± 33.8 253.2 ± 29.1	8.9 ± 1.3 7.8 ± 1.7	0.56 ± 0.08 0.48 ± 0.11	This work
	(0.25% H ₂ SO ₄ v/v, 121°C, 21 psi, 1h)	31°C	1 10	$\begin{array}{c} 5574.8 \pm 156.6 \\ 5063.0 \pm 233.5 \end{array}$	350.6 ± 9.8 342.7 ± 14.3	7.9 ± 0.6 7.9 ± 0.3	$\begin{array}{c} 0.50 \pm 0.04 \\ 0.48 \pm 0.01 \end{array}$	

^a Converted unit from the original data







Fig. 8.



586 Fig. 9.



Fig. 10.



Fig. 11. 591



593 Fig. 12.