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Fermentation of lactose and its constituent sugars by *E. coli* WDHL: Impact on hydrogen production

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1 **Abstract**

2 Fermentations of lactose, glucose and galactose using *Escherichia coli* WDHL, a hydrogen
3 overproducer strain, were performed. The results showed that the pyruvate is mainly routed
4 to the lactate pathway using glucose. Thus the formate and consequently hydrogen
5 production ~~was~~ were diminished. The hydrogen production and yield obtained with glucose
6 were 1037 mL and **0.30** mol H₂/mol of glucose, respectively. The galactose catabolism was
7 slower than ~~the~~ that of glucose ~~one~~. Using galactose, the pyruvate formate lyase pathway
8 was the main route for pyruvate; the ethanol production was also favored. The galactose
9 fermentation yield **1.12** mol H₂/mol of galactose and the hydrogen production was 2080
10 mL. The fermentation of lactose or glucose plus galactose showed similar behavior.
11 Lactose yield was 1.02 mol H₂/mol of lactose. This work provides valuable information
12 which can be used for the improvement of hydrogen production using lactose, glucose or
13 galactose rich wastes.

14

15 **Keywords: glucose, galactose, lactate, hydrogen yield, *Escherichia coli* WDHL**

16

17 **1. Introduction.**

18 ~~The biofuels~~Biofuel production is a very active research area due to the future depletion of
19 fossil fuels and the environmental problems associated with ~~the use of them~~their use
20 (Luque et al., 2008). ~~-Among the~~ biofuels, biohydrogen is an attractive future substitute ~~of~~
21 for fossil fuels due to its potentially higher efficiency of conversion to usable power, low or
22 non-generation of pollutants and high energy density (Antonopoulou et al., 2011;
23 Hallenbeck and Ghosh, 2009; Sinha and Pandey, 2011). ~~The b~~Biological hydrogen
24 production is carried out at ambient temperature and pressure, ~~by this reason; therefore it is~~
25 less energy intensive than ~~the~~ chemical and electrochemical processes (Nath and Das, 2004;
26 Rosales-Colunga et al., 2010a). ~~But~~ However, to be competitive ~~to the~~with other methods
27 of production, ~~the~~ biological production must use ~~wastes~~ waste or by-products rich in
28 carbohydrates, ~~thus and then~~ reducing the cost of production and, at the same time,
29 disposing ~~off of~~ pollutant wastes.

30 ~~The dark~~Dark fermentation is a promising biological process to obtain hydrogen because it
31 could use organic ~~wastes~~ waste from ~~the~~ agricultural and food-producing ~~industry~~
32 industries as substrates (Liu et al., 2008), and only ~~a~~ relatively simple equipment is
33 necessary (Hallenbeck, 2005). *Escherichia coli* produce hydrogen by dark fermentation
34 under anaerobic conditions when no external electron acceptors are present
35 (Leonhartsberger et al., 2002; Sawers, 2005). The hydrogen yield can be improved by
36 genetic engineering, and *E. coli* is one of the microorganisms most used because ~~of its~~
37 genetics and metabolism are well documented (Davila-Vazquez et al., 2008a). The
38 ~~substrate more~~most extensively studied substrate for hydrogen production ~~had~~ has been
39 glucose (Maeda et al., 2007; Maeda et al., 2008; Penfold et al., 2003; Yoshida et al., 2006).

40 | Lactose, and the sugars released from its hydrolysis: ~~glucose~~ and galactose, are commonly
41 | present in some agro-industrial ~~wasteswaste~~. Lactose is found in ~~the~~ cheese and dairy
42 | industry wastewater (Calli et al., 2008; Chong et al., 2009). Glucose is currently obtained
43 | from the hydrolysis of molasses, cellulose, and other agricultural wastes (Kapdan and
44 | Kargi, 2006). Besides the release of galactose from lactose hydrolysis, this sugar is a
45 | component of hemicellulose (Ren et al., 2009). Thus, it is interesting to study the use of
46 | these carbohydrates as single substrates or in mixture for hydrogen production. Despite ~~the~~
47 | ~~fact that~~ the use of glucose by *E. coli* has been extensively studied as mentioned above, the
48 | use of lactose and galactose as substrates ~~have~~~~has~~ not been sufficiently studied.

49 | Consequently, the fermentation of lactose, glucose, and galactose using *Escherichia coli*
50 | WDHL was studied; in this strain the *hycA* and *lacI* genes were deleted to improve ~~the~~
51 | hydrogen production (Rosales-Colunga et al., 2010b). The *hycA* ~~gen~~~~gene~~ codes for the
52 | negative regulator of the hydrogen pathway whereas the deletion of *lacI* lead to the
53 | constitutive expression of the lac operon.

54 |

55 | **2. Materials and methods**

56 | **2.1 Strain and culture media.**

57 | *Escherichia coli* WDHL strain, a hydrogen ~~over producer~~~~overproducer~~ strain, which lacks
58 | *hycA* and *lacI* genes, was used. A complete description of the strain has been published
59 | (Rosales-Colunga et al., 2010b). ~~Preinoecula~~~~Preinnocula~~ were grown overnight in 25 mL of
60 | LB medium at 37°C and shaken at 200 rpm, afterwards added to 900 mL of fresh LB
61 | medium in closed twist cover bottles and were incubated at 37°C for 48 h. Cells were

62 | harvested, washed₂ and inoculated into the bioreactor. Cultures on the bioreactor were done
63 | using HP medium reported elsewhere (Rosales-Colunga et al., 2010b) with 15 g/L of sugars
64 | (lactose, glucose, galactose or a mixture of 7.5 g/L of glucose and 7.5 g/L of galactose). A
65 | duplicate using glucose in M9 medium was ~~done-made~~ to validate the reproducibility of the
66 | experiments.

67

68 | **2.2 Batch Cultures.**

69 | Cultures were performed in batch mode using a 1-L bioreactor (Applikon, Schiedam, The
70 | Netherlands). The pH, oxidation-reduction potential, dissolved oxygen and dissolved
71 | carbon dioxide were monitored using autoclavable electrodes (Applikon) connected to the
72 | ADI 1035 Bioconsole (Applikon). The initial pH was 7.5 in all the experiments and then
73 | automatically controlled to 6 using 2.5 N NaOH and HCl solutions. BioXpert 1.3 software
74 | (Applikon) was used for data acquisition. The cultures were maintained at 37°C and stirred
75 | at 175 rpm with two six-blade Rushton turbines.

76

77 | **2.3 Analytical methods**

78 | Cell growth was monitored at OD_{600nm} using a Cary BIO-50 spectrophotometer ~~Cary BIO-~~
79 | ~~50~~(Varian, Palo Alto, CA). Culture samples were periodically taken from the bioreactor,
80 | and centrifuged₂ and the supernatant was filtered through a 0.22 µm filter (Millipore) for
81 | the analysis of sugars, organic acids and ethanol. The gas produced was measured by water
82 | displacement in an inverted burette connected to the bioreactor with rubber tubing and a

83 needle. The hydrogen content in the gas phase, sugars and organic acids were determined
84 by gas chromatography and capillary electrophoresis as described elsewhere (Davila-
85 Vazquez et al., 2008b). Ethanol was determined by gas chromatography as described
86 elsewhere (De Leon-Rodriguez et al., 2006).

87

88 3. Results

89 3.1 Fermentation of lactose

90 Lactose is commonly present in ~~wastes of~~ food industry waste, and it can be used as
91 substrate for the production of hydrogen (Chong et al., 2009; Kapdan and Kargi, 2006;
92 Keskin et al.). A typical batch culture using lactose as substrate is shown in Fig. 1. ~~It can be~~
93 ~~noted~~ Three phases of hydrogen production can be noted (Fig. 1A). In the first 60 h the
94 hydrogen specific production rate was 0.34 mmol H₂/L h OD₆₀₀unit and 980 ml were
95 produced. In the next following 35 h, the hydrogen production rate was 0.02 mmol H₂/L h
96 OD₆₀₀unit, and only 35 ml were produced. After 95 h and onwards, hydrogen was
97 continuously produced reaching 2092 ml in 576 h. The maximum hydrogen production rate
98 was 15.41 mL/L h.

99 When lactose was consumed, galactose and glucose were accumulated. Glucose reached a
100 maximum of 3.2 g/L at 9 h and became undetectable after 34 h. The maximum galactose
101 concentration was 7.45 g/L at ~~the~~ 27 h, after ~~that~~ which the concentration decreased to 1.2
102 g/L at the end of the experiment. Lactose was not detected after 12 h (Fig. 1B).

103 The soluble metabolites produced are shown in Fig. 1C. Lactate was the main soluble
104 product and its maximum concentration was 5 g/L at 27 h and remained constant until the

105 end of the experiment. The production of acetate and ethanol showed a similar profile and
106 two phases were observed for both metabolites. In the first phase, a rapid ~~increment~~
107 increase was observed. ~~A~~ Acetate reached 1 g/L and ethanol 0.7 g/L at 27 h, after ~~that~~
108 which the concentration of both metabolites steadily increased to a maximum concentration
109 of 2.6 and 2.5 g/L, respectively. ~~The~~ Succinate production also showed two phases, 0.4 g/l
110 were produced in the first 27 h and the maximum concentration was 1 g/L. Only a slight
111 amount of formate was detected on the culture medium. Formate reached a maximum of
112 0.2 g/L at 12 h, and it was undetectable at 70 h.

113 It is clear that after the lactose was hydrolyzed, it was consumed in two phases. In the first
114 one the glucose was consumed and galactose was accumulated, in the second phase
115 galactose was slowly used and the metabolites produced were different in the two phases.
116 In order to study both phases, independent experiments using either glucose or galactose as
117 substrates were conducted.

118

119 **3.2 Fermentation of glucose.**

120 A typical batch culture using glucose as substrate is shown in Fig. 2. As expected, this
121 carbohydrate was used immediately and was not detected after 44 h. ~~The~~
122 ~~hydrogen~~ Hydrogen production also started ~~since from~~ the beginning of the culture. ~~T~~
123 the final hydrogen production was 1037 mL in mainly in 60 h (Fig. 2A). In this case the
124 maximum hydrogen production rate was 18.61 mL/L h. ~~The~~ ~~d~~ Duplicate attained a final
125 production and a maximum hydrogen production rate of 965 mL and 15.3 mL/L h,
126 respectively.

127 The production of metabolites is shown in ~~the~~ Fig. 2B. Lactate was the main soluble
128 product of the fermentation and attained a maximum concentration of 10.1 g/L₂ whereas the
129 other products were produced in ~~a~~ minor concentration. The maximum concentration_s of
130 succinate, acetate, and ethanol were 1.6, 1.5 and 1.2 g/L, respectively. The highest formate
131 concentration ~~was reached~~ 0.6 g/L at 12 h and then decreased.

132

133 3.3 Fermentation of galactose.

134 A typical batch culture using galactose as substrate is shown in Fig. 3. In this case, a lag
135 phase of 18 h was observed. ~~Other~~ Another marked difference was the ~~time for the~~
136 galactose up-take time; whereas the glucose was completely consumed at 44 h (Fig. 2A),
137 galactose required a longer culture time and, at 356 h of culture, 4.2 g/L of galactose still
138 remained in the culture medium (Fig. 3A). Since the galactose consumption became
139 asymptotic, the culture was stopped. The hydrogen production from galactose is ~~showed~~
140 shown in Fig. 3A. ~~Due~~ Due to the lag phase in the galactose consumption the hydrogen
141 production began 18 h after the experiment started. The hydrogen production attained 2080
142 mL in 356 h. This production represents ~~two-fold~~ two times the hydrogen produced from
143 glucose despite no total galactose was consumed. The maximum hydrogen production rate
144 (13.21 mL/L h) using galactose was lower than when using glucose.

145 The production of metabolites from galactose is presented in Fig. 3B. In this case, the main
146 soluble metabolite was ethanol with a final concentration of 6.1 g/L. Acetate and succinate
147 were also produced and reached 2.7 and 1.7 g/L, respectively. In contrast, with the

148 fermentation of glucose, galactose produced only 0.4 g/L of lactate. The formate
149 concentration was less than 0.2 g/L during the fermentation.

150

151 **3.4 Fermentation of a mixture of glucose and galactose.**

152 To investigate the differences observed in the production of hydrogen and other metabolites
153 when glucose or galactose were used as single substrates, an experiment using glucose plus
154 galactose was carried out. ~~The~~ Fig. 4 shows the sugars consumption and production of
155 hydrogen and soluble metabolites. This culture clearly shows two phases. In the first phase
156 glucose was quickly consumed in the initial 25 h, and ~~the~~ hydrogen production reached 983
157 mL (Fig. 4A). Lactate was the main soluble product in this phase and attained a
158 concentration of 6.7 g/L at 25 h. The concentrations of succinate, acetate, and ethanol in the
159 first 25 h were 0.5, 1.5 and 1.2 g/L, respectively. Formate peaked at 19 h with a
160 concentration of 0.5 g/L and then decreased (Fig. 4B). After glucose was depleted, a lag
161 phase of 120 h was necessary to start the galactose consumption. In this case the lag-phase
162 was 6.7-times higher than the culture started with galactose. Interestingly, galactose was
163 consumed completely at 320 h of culture and 1467 mL of hydrogen were produced, which
164 is nearly 50% more hydrogen than ~~the~~ that produced from ~~the~~ glucose (Fig. 4A). Acetate,
165 ethanol, and succinate were produced in this second phase and reached a maximum
166 concentration of 4, 4, and 1.6, respectively, whereas lactate remained constant. Formate
167 also showed a peak in its concentration and reached 0.5 g/L at 200 h, and then decreased
168 and was undetectable at the end of fermentation. The maximum hydrogen production rate
169 was 24.45 mL/L h during this experiment.

170

171 3.5 Comparison of hydrogen production.

172 Measures ~~on~~of hydrogen and soluble metabolites produced by the fermentation of lactose,
173 glucose and galactose showed differences. Table 1 shows the differences ~~on~~in hydrogen
174 production. Using glucose as substrate resulted ed in low hydrogen production and poor yield
175 (0.30 mol H₂/mol of glucose), with the main soluble product was being lactate. ~~Whereas~~
176 whereas galactose gave the highest hydrogen yield (1.12 mol H₂/mol galactose) and the
177 main product was ethanol. Lactate was the main product using lactose, but in this case ~~the~~
178 lactate was produced only when the glucose was consumed. Interestingly in the
179 fermentation of glucose plus galactose, the highest production of hydrogen was attained,
180 and the main products were lactate, acetate and ethanol. The yield of this fermentation was
181 the same as that of the ~~lactose one~~ (1.02 mol H₂/mol hexose). The maximum hydrogen
182 production rate using glucose plus galactose of 24.45 ml/L h was higher than the 15.41
183 ml/L h attained with lactose. Galactose yielded the lowest maximum production rate.

184

185 4. Discussion

186 Although the fermentation of glucose to hydrogen is straightforward, the main drawback is
187 that only a small fraction of the electrons in the starting substrate ends up in hydrogen
188 (Rittmann, 2008). In *E. coli*, glucose is transported into the cell by the phosphotransferase
189 system and then catabolized to phosphoenolpyruvate. ~~Tand~~ and this is the first branch of the
190 fermentative pathway because it can be converted to oxaloacetate, and at last produces
191 succinate. In the other branch, most of the phosphoenolpyruvate is transformed ~~to~~into

192 pyruvate, which is cleaved to formate and acetyl-CoA by pyruvate formate lyase complex.
193 The formate is converted to hydrogen and CO₂, whereas the latter yields acetate or ethanol
194 (Clark, 1989). ~~But~~However, during-under conditions of high pyruvate accumulation or at
195 low pH, pyruvate may be converted to lactate by lactate dehydrogenase enzyme (LDH)
196 coded by the *ldhA* gene (Tarmy and Kaplan, 1968). The glucose uptake rate was 210.83 mg
197 of glucose/L h OD₆₀₀ unit, ~~and that means meaning~~ that the glucose is quickly converted to
198 pyruvate, ~~and~~ and therefore the concentration of pyruvate must be high. Then, the lactate
199 pathway must be very active since the LDH activity increases with increased pyruvate
200 concentration (Tarmy and Kaplan, 1968) and it has been showed that the addition of
201 exogenous pyruvate increased the expression of *ldhA* (Jiang et al., 2001), and this might be
202 the reason why lactate was the main soluble metabolite and the hydrogen yield was low
203 using glucose.

204 The hydrogen yield from glucose found in the present work, 0.30 mol H₂/mol of glucose, is
205 higher than the yield of 0.17 mol H₂/mol of glucose ~~consumed-consumption~~ predicted by
206 metabolic flux analysis reported by- Manish et al., 2007. They also predicted an ~~increment~~
207 increase of 35% in hydrogen yield in a strain lacking the *ldhA* gene. Other studies that
208 found higher yield include -,Bisaillon et al., 2006, which reported the highest yield of 2
209 mol H₂/mol of glucose, using a strain with mutations on uptake hydrogenases, *ldhA* and
210 *fhlA*, in batch cultures and limiting concentrations of glucose. Similar yields were reported
211 by Turcot et al., 2008, with the same strain and nutrient limitations in continuous cultures,
212 whereas Ghosh and Hallenbeck, 2009 attained 1.51 mol H₂/mol of glucose using the same
213 strain. Maeda et al., 2007, reached a hydrogen yield of 1.3 mol H₂/mol of glucose using a
214 strain with mutations on *hyaB hybC hycA fdoG frdC ldhA aceE* genes. Yoshida et al., 2006

215 | enhanced the hydrogen yield to 1.82 mol H₂/mol of glucose with a strain of *ΔldhA*,
216 | *ΔfrdBC*. Mathews et al., 2010 obtained the same hydrogen yield with the strain GW16
217 | (*ΔhyaAB, ΔhybABC, ΔhycA, ΔfrdBC ΔldhA*) using rich defined media. They also
218 | observed an increment in the acetate concentration. In the results presented here the acetate
219 | and hydrogen production were low. It is important to notice that in all the works discussed
220 | above a mutation in the *ldhA* gene was included, and the lactate pathway abolished,
221 | contributing to the increase ~~the in~~ hydrogen yield. ~~Other-Another~~ study performed by
222 | Penfold et al., 2003 did not involve a mutant of *ldhA*. They used a *hycA* mutant strain and
223 | ~~they~~ found that the amount of hydrogen decreased as the concentration of glucose
224 | increased. It could be that ~~the~~ lactate pathway is not very active due to the low pyruvate
225 | concentration caused by the low glucose concentration used.

226 | The galactose is important for *E. coli* not only as an energy source but also as a building
227 | block in complex polysaccharide synthesis. The transport of galactose, unlike that of
228 | glucose, is by two specific transporting systems, one of high affinity and one of low
229 | affinity, but it can also be transported by LacY permease and other non-specific
230 | transporters (Weickert and Adhya, 1993). In the galactose fermentations a lag phase is
231 | observed because the *gal* operon is not activated immediately despite the high galactose
232 | concentration and the constitutive presence of LacY permease due to the lack of *lacI* gene
233 | in the strain used here. After this lag phase galactose catabolism began but the galactose
234 | consumption rate was slower than that of glucose (26.47 mg of galactose/L h OD₆₀₀ unit).
235 | Thus the pyruvate concentration is low, the lactate pathway is poorly activated and the
236 | hydrogen pathway is strongly favored. I was also suggested ~~Also it was suggested~~ that the
237 | expression of the *ldhA* gene might be affected by the nature of carbon source, and ~~with by~~

238 the PTS system (Jiang et al., 2001). The difference on the uptake rates between glucose and
239 galactose must be due to the transportation or the enzymes related to the metabolism of
240 galactose before it can be converted to glucose 6-P. If the galactose transportation is caused
241 the slow consumption, then the galactose can be metabolized quickly in the fermentation of
242 lactose. Since, ~~in the case of~~ lactose, which is transported by the lactose permease and then
243 intracellular lactose is split into glucose and galactose by β -galactosidase (Kremling et al.,
244 2007), the transportation of galactose is not involved. ~~Since, in~~ this case the galactose was
245 slowly consumed suggesting that it is due to the inefficient expression of *gal* regulon or low
246 activity of the enzymes coded by this regulon. Finally, in the experiment of glucose plus
247 galactose as substrates, the lag phase for galactose uptake is longer, which-and-it could be
248 caused by catabolic repression, since both glucose and galactose were present in the culture
249 medium (Adhya and Echols, 1966). The hydrogen yields from galactose and lactose were
250 1.12 and 1.02 mol H₂/mol hexose consumed using a strain lacking the *hycA* and *lacI* genes.
251 These yields are higher than the 0.69 and 0.73 mol H₂/mol reported by Ghosh and
252 Hallenbeck, 2009, for galactose and lactose, respectively using a strain with mutations on
253 uptake hydrogenases, *ldhA* and *fhIA*. It seems that the mutation on *lacI* improves s the
254 hydrogen yield from those sugars.

255

256 **5. Conclusions**

257 The two phases of metabolites production using lactose are due to the consumption of
258 glucose and after that, ~~the~~ galactose consumption. Using glucose, pyruvate is mainly
259 channeled to the lactate pathway and hydrogen production is diminished. Using galactose,

260 the formate and acetyl Co-A pathway is the main route, hydrogen and ethanol are the main
261 products. The maximum hydrogen production rate is high when glucose is present, whereas
262 galactose yields the lowest maximum production rate. The presence of glucose in the
263 culture medium produces a longer lag phase of galactose than the lag phase of galactose as
264 a sole carbon source.

265

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271

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373 **Figure captions:**

374 **Fig. 1.** Batch culture of *E. coli* WDHL using lactose as substrate. **A** Hydrogen production
375 (■). **B** Sugars consumption: lactose (✦), galactose (●) and glucose (△). **C** Production of
376 fermentative metabolites: succinate (+), lactate (□), formate (●), ethanol (△) and acetate
377 (▲).

378 **Fig. 2.** Batch culture of *E. coli* WDHL using glucose as substrate. **A** Hydrogen production
379 (■) and glucose consumption (✦) **B** Production of fermentative metabolites: succinate
380 (+), lactate (□), formate (●), ethanol (△) and acetate (-▲-).

381 **Fig. 3.** Batch culture of *E. coli* WDHL using galactose as substrate. **A** Hydrogen production
382 (■) and galactose consumption (✦) **B** Production of fermentative metabolites: succinate
383 (+), lactate (□), formate (●), ethanol (△) and acetate (-▲-).

384 **Fig. 4.** Batch culture of *E. coli* WDHL using a mix of glucose plus galactose as substrate.
385 **A.** Hydrogen production (■) and sugars consumption: glucose (△), galactose (●) **B.**
386 Production of fermentative metabolites: succinate (+), lactate (□), formate (●), ethanol
387 (△) and acetate (-▲-).

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