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1 **Environmental Context for the Terminal Ediacaran Biomineralization of Animals**

2
3 Short running title: **Environmental context for animal biomineralization**

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23
24 **ABSTRACT**

25 In terminal Ediacaran strata of South China, the onset of calcareous biomineralization is preserved in
26 the paleontological transition from *Conotubus* to *Cloudina* in repetitious limestone facies of the Dengying
27 Formation. Both fossils have similar size, funnel-in-funnel construction, and epibenthic lifestyle,
28 but *Cloudina* is biomineralized whereas *Conotubus* is not. To provide environmental context for this
29 evolutionary milestone, we conducted a high-resolution elemental and stable isotope study of the richly
30 fossiliferous Gaojiashan Member. Coincident with the first appearance of *Cloudina* is a significant positive
31 carbonate carbon isotope excursion (up to +6‰) and an increase in the abundance and ³⁴S composition of
32 pyrite. In contrast, δ³⁴S values of carbonate-associated sulfate (CAS) remain steady throughout the succession,
33 resulting in anomalously large (>70‰) sulfur isotope fractionations in the lower half of the member. The
34 fractionation trend likely relates to changes in microbial communities, with sulfur disproportionation involved
35 in the lower interval whereas microbial sulfate reduction was the principal metabolic pathway in the upper. We
36 speculate that the coupled paleontological and biogeochemical anomalies may have coincided with an increase
37 in terrestrial weathering fluxes of sulfate, alkalinity, and nutrients to the depositional basin, which stimulated
38 primary productivity, the spread of an oxygen minimum zone, and the development of euxinic conditions in
39 subtidal and basinal environments. Enhanced production and burial of organic matter is thus directly connected
40 to the carbon isotope anomaly, and likely promoted pyritization as the main taphonomic pathway for
41 *Conotubus* and other soft-bodied Ediacara biotas. Our studies suggest that the Ediacaran confluence of
42 ecological pressures from predation and environmental pressures from an increase in seawater alkalinity set the
43 stage for an unprecedented geobiological response: the evolutionary novelty of animal biomineralization.

44
45 **INTRODUCTION**

46 One of the earliest animals to have developed a biomineralized carbonate exoskeleton is *Cloudina* —
47 named after the famed Precambrian paleontologist Preston Cloud (1912–1991) and preserved in terminal
48 Ediacaran (ca. 550–541 Ma) sedimentary successions worldwide (Conway Morris et al., 1990; Sour-Tovar et
49 al., 2007; Gaucher and Germs, 2009; Cortijo et al., 2010; Zhuravlev et al., 2012). This animal, which is
50 suggested to be an ancient cnidarian-grade (Grant, 1992; Cortijo et al., 2010) or lophotrochozoan animal (Hua
51 et al., 2005; Zhuravlev et al., 2015), constructed a high-Mg calcitic tubular shell with nested funnels, had an
52 epibenthic lifestyle with its apex attached to the substrate (Grant, 1990; Zhuravlev et al., 2012; Cai et al., 2014),
53 and may have had both sexual and asexual reproductive strategies to aid in its broad ecological dispersal

(Cortijo et al., 2015). *Cloudina* was associated with microbial reefs, and may have been a reef builder like modern-day corals that inhabit oligotrophic shelf environments where they band together in search of hard substrates and for protection against predators (Penny et al., 2014; Wood and Curtis, 2015).

Biomineralization of *Cloudina* is widely considered to have been a response to predation given the significant number of borings found on its fossil shells (Bengtson and Zhao, 1992; Hua et al., 2003; Porter, 2011). In addition, some also have considered terminal Ediacaran biomineralization as a (toxico-)physiological response to regulate calcium concentrations in circulatory fluids (Simkiss, 1977; Kempe et al., 1989; Simkiss, 1989; Brennan et al., 2004), or to environmental perturbations involving oscillations in atmospheric $p\text{CO}_2$ and seawater chemistry (Knoll, 2003a; Knoll and Fischer, 2011). Environmental drivers, however, are particularly difficult to assess insofar as these should also have a broad effect on general biotic diversification and vice versa (Knoll, 2003b; Gaidos et al., 2007; Butterfield, 2009; Butterfield, 2011; Lenton et al., 2014; Erwin, 2015). Insofar as there is a metabolic cost to biomineralization, the biological benefits to the organisms, including protection against predation and the physiological response of organisms to rapidly changing seawater chemistry in the terminal Ediacaran Period, should be balanced (Knoll, 2003a; Xiao, 2014).

To this end, we investigated a Lagerstätte of Ediacaran animals in the Gaojiashan Member of the Dengying Formation in South China (**Fig. 1**) (Hua et al., 2007; Cai et al., 2010). Within the member's repetitious limestone facies, the first appearance of *Cloudina* is immediately preceded by *Conotubus*, a soft-bodied antecedent exquisitely preserved through pyritization with similar construction, size, and lifestyle (Cai et al., 2011; Cai et al., 2014). To explore this unique paleontological juxtaposition and provide environmental context for the earliest examples of animal biomineralization and pyritization, we sampled the Gaojiashan Member at high stratigraphic resolution for elemental and isotopic compositions. Our chemostratigraphic investigation reveals that these evolutionary and taphonomic events are associated with profound biogeochemical shifts in both the carbon and sulfur cycles, and speculate that they may have been facilitated by profound environmental perturbations in the marine realm ultimately driven by enhanced oxidative weathering of the continents.

METHODS

In this study, the Gaojiashan Member was systematically sampled at high resolution for integrated chemostratigraphic and geochronological investigations. Geochemical analyses were conducted in the Paleoclimate CoLaboratory at University of Maryland. Details of the methods used in the CoLaboratory can be found in previous publications (e.g. McFadden et al., 2008; Zhelezinskaia et al., 2014; Cui et al., 2015), but are briefly outlined below.

Carbon and oxygen isotope analysis

Rock samples were cut and polished for detailed petrographic observation and micro-drilling in order to obtain powders from the least-altered, least-recrystallized, and purest phases for carbonate carbon ($\delta^{13}\text{C}_{\text{carb}}$) and oxygen ($\delta^{18}\text{O}_{\text{carb}}$) isotope analysis. The powders were measured with a Multicarb inlet device in-line with an Elementar Isoprime continuous-flow isotope ratio mass spectrometer, and precision for both isotopes was routinely better than 0.1‰.

Elemental analyses

Major and trace elemental abundances in carbonates were analyzed in order to better evaluate the degree of diagenetic alteration. Aliquots of the micro-drilled carbonate powders were dissolved in 0.4 M HNO_3 , centrifuged, and only analyzed for the solutions. Petrographic observations indicate that these powders were largely free of siliciclastics; any clays, if present, would not have been dissolved by the dilute acid. The resulting solutions were analyzed on a Thermo Scientific® iCAP-Q ICP-MS (Inductively Coupled Plasma – Mass Spectrometry) at the Carnegie Institution of Washington. Precision of these analyses as determined by repeated measurements of a house standard carbonate was < 5% (2σ) for major elements with high concentrations and < 10% (2σ) for the REEs.

Organic carbon and paired sulfur isotope analyses

107 The organic carbon ($\delta^{13}\text{C}_{\text{org}}$), total sulfur ($\delta^{34}\text{S}_{\text{TS}}$ of pyrite and trace amount of organic S) isotope
108 compositions were measured by combustion of the decalcified residuals to CO_2 or SO_2 with a Eurovector
109 elemental analyzer in-line with a second Elementar Isoprime isotope ratio mass spectrometer. Bulk carbonate
110 powders were used for extraction of carbonate-associated sulfate (CAS). For the former, ~15 g of bulk crushed
111 sample was acidified with 3 M HCl. These acidified residues were washed with ultra-pure Milli-Q (18M Ω)
112 water, centrifuged, decanted, and dried. For the latter, ~100 g of crushed bulk sample, which were repeatedly
113 leached with 10% NaCl solutions. To minimize the contamination of soluble non-CAS sulfate (Marenco et al.,
114 2008; Wotte et al., 2012; Peng et al., 2014; Schobben et al., 2015), bulk powders were leached by 10% NaCl
115 solutions for at least 10 times with at least two hours for each time, and then washed with Milli-Q water for at
116 least 3 times prior to acidification of the leached powders with 3 M HCl. CAS precipitates were then collected
117 as BaSO_4 three days after BaCl_2 was added to the solution. The residues and the BaSO_4 precipitates were
118 packed into folded tin cups with V_2O_5 for combustion to CO_2 or SO_2 in a Eurovector elemental analyzer in-line
119 with a second Elementar Isoprime isotope ratio mass spectrometer, which measured isotope abundances.
120 Uncertainties for carbon and sulfur isotope measurements determined by multiple analyses of standard
121 materials during analytical sessions are better than 0.1% and 0.3‰, respectively.
122

123 *Detrital zircon dating*

124 For all aspects of zircon dating we followed the procedures described in Martin et al. (2015). Zircon
125 grains were isolated using conventional mineral separation techniques including rock pulverization by hand
126 using a mortar and pestle, removal of silt and clay by hand panning in water, removal of magnetic grains using
127 a Frantz magnetic barrier separator, and density separation using methylene iodide. Zircon grains were then
128 poured onto double-sided tape and cast them in an epoxy disk along with approximately 10 shards of the Sri
129 Lanka zircon standard (564 \pm 3 Ma) (Gehrels et al., 2008). After hand polishing to expose the interiors of the
130 grains, we produced backscattered electron and cathodoluminescence images using the JEOL JXA-8900R
131 electron probe microanalyzer at the University of Maryland.
132

133 The cores of 49 and 175 zircon grains were dated from samples 09G-35.3 and 09G-37.9, respectively,
134 by laser ablation–inductively coupled plasma–mass spectrometry in the Arizona LaserChron Center at the
135 University of Arizona, taking care to avoid multiple cathodoluminescence zones, inclusions, and cracks.
136 Ablation of the zircon was performed using a New Wave UP193HE Excimer laser and a spot diameter of 30
137 μm . The ablated zircon was carried in helium into the plasma source of a Nu Plasma HR multi-collector mass
138 spectrometer, and analyses followed the protocols described in Martin et al. (2015).
139

140 Corrections for inter-element fractionation of Pb/U and common Pb, as well as other data reduction,
141 were performed off-line using an Excel program developed at the Arizona LaserChron Center. We removed
142 from further consideration analyses with: (1) high ^{204}Pb , (2) greater than 5% error on the $^{206}\text{Pb}/^{207}\text{Pb}$ date, (3)
143 greater than 5% error on the $^{206}\text{Pb}/^{238}\text{U}$ date, (4) greater than 25% normal discordance or 8% reverse
144 discordance, (5) high U concentration, or (6) high U/Th ratio. The remaining analyses were used in our
145 interpretations (**Table S1; Fig. 3**). Isoplot was used to calculate weighted means and to produce concordia and
146 probability density plots (Ludwig, 2008).
147

148 $^{206}\text{Pb}/^{238}\text{U}$ dates are usually more precise than $^{206}\text{Pb}/^{207}\text{Pb}$ dates for zircon younger than about 1.4 Ga,
149 whereas the reverse is true for older grains. However, $^{206}\text{Pb}/^{207}\text{Pb}$ dates are only minimally affected by recent
150 lead loss, so in most cases they more closely indicate the time of crystallization for zircon older than about 1
151 Ga. Therefore, during interpretation we used $^{206}\text{Pb}/^{238}\text{U}$ dates for grains younger than 1 Ga and $^{206}\text{Pb}/^{207}\text{Pb}$
152 dates for older zircon grains.
153

154 **STRATIGRAPHIC AND GEOCHRONOLOGICAL CONSTRAINTS**

155 Carbonates of the Dengying platform are sandwiched between the Ediacaran Doushantuo Formation
156 (ca. 635-551 Ma) (Jiang et al., 2011) and the early Cambrian Kuanchuanpu Formation (Steiner et al., 2004) in
157 the southern Shaanxi region (**Fig. 1**). The Dengying Formation at Gaojiashan is subdivided into three intervals,
158 including the Algal Dolomite, Gaojiashan, and Beiwan members (**Fig. 1C**), which are generally correlated
159 with Hamajing, Shibantan, and Baimatuo members, respectively, in the Yangtze Gorges area (Zhou and Xiao,

160 2007; Zhu et al., 2007; Duda et al., 2015). Based on the 551 Ma U-Pb zircon depositional age of a volcanic ash
161 at the top of the Miaohe Member, which has historically been correlated with Doushantuo Member IV
162 (Condon et al., 2005), and an estimated 541 Ma age for the Ediacaran-Cambrian boundary (Amthor et al., 2003;
163 Chen et al., 2015), the >650 m thick Dengying Formation represents the last 10 million years of the Ediacaran
164 Period. However, a recent chemostratigraphic study of the Miaohe Member (An et al., 2015) demonstrates that
165 the 551 Ma ash bed lies between the Hamajing and Shibantan members of the Dengying Formation, and is thus
166 not relevant to the biogeochemical anomaly (i.e. Shuram Excursion) preserved in the upper Doushantuo
167 Formation (*cf.* Kaufman, 2005).

168
169 At the studied section, the Gaojiashan Member is 55 m in thickness, including a siltstone interval in
170 the lower part, repetitious siltstone-mudstone-limestone facies with cryptalgal crinkly laminations in the
171 middle part, and a coarse sandstone/conglomerate at the top (**Fig. 2**) (Cai et al., 2010). The lower Gaojiashan
172 Member contains the enigmatic fossil *Shaanxilithes ningqiangensis* preserved in siltstone facies (Meyer et al.,
173 2012). The middle Gaojiashan Member contains *Conotubus hemiannulatus* and *Gaojiashania cyclus* preserved
174 in thin, normally graded calcisiltite-siltstone beds interpreted as distal event deposits (Cai et al., 2010),
175 followed by the first appearance of the biomineralized animal *Cloudina* preserved in intraclastic limestones
176 approximately 40 m above the base of the succession (**Fig. 1**). A distinctive horizon with bedded gypsum
177 occurs in the upper part of the Gaojiashan Member (**Figs. 1D, 2F-G**).

178 179 RESULTS

180 Detrital zircon from two closely-spaced siltstone beds at 16.7 m (sample 09G-35.3) and 14.1 m
181 (sample 09G-37.9) above the base of the Gaojiashan member yielded a youngest population of four U-Pb ages
182 ranging from 543 to 550 Ma from sample 09G-37.9, with a weighted mean age of 548 ± 8 Ma (MSWD = 0.11)
183 (**Fig. 3, Table S1**). This maximum depositional age based on detrital zircon ages is consistent with the 551 Ma
184 age estimate for the Miaohe Member beneath the Gaojiashan equivalent Shibantan Member of the Dengying
185 Formation based on U-Pb zircon age from the bedded ash layer (Amthor et al., 2003; Condon et al., 2005;
186 Chen et al., 2015). However, most of the detrital zircons in this study had ages between 750 and 850 Ma, with
187 a scattering of solitary dates spanning from 1300 to 2700 Ma.

188
189 In total 113 limestone and calcareous siltstone samples from the Gaojiashan Member were analyzed
190 for elemental abundances and isotopic compositions (**Figs. 4-5; Tables S2-S4**). Carbonate percentages in the
191 samples are generally high (>90%), except in the lower member where limestones and siltstones are
192 interbedded. The stratigraphic trend of $\delta^{13}\text{C}_{\text{carb}}$ variations reveals a positive carbon isotope excursion (up to
193 +6‰) in the upper part of the Gaojiashan Member, coinciding with the fossil transition from *Conotubus* to
194 *Cloudina*. Coupled with the positive $\delta^{13}\text{C}_{\text{carb}}$ event, $\delta^{13}\text{C}_{\text{org}}$ data reveal a negative excursion (down to -30‰);
195 calculated carbon isotope fractionations ($\Delta\delta^{13}\text{C}_{\text{carb-org}}$) show peak values up to +36‰ in this interval. Pyrite S
196 isotope ($\delta^{34}\text{S}_{\text{pyrite}}$) values measured from bulk acidified residues (assuming pyrite S \gg organic S) show a wide
197 range from -30‰ to +30‰ in the Gaojiashan Member, with more negative values in the lower half of the
198 section, and more positive values in the upper half. On the contrary, sulfur isotope compositions of carbonate-
199 associated sulfate ($\delta^{34}\text{S}_{\text{CAS}}$) remain generally invariant around ca. +40‰ throughout the Gaojiashan Member.
200 Both total sulfur (TS) and total organic carbon (TOC) are relatively low throughout most of the succession, but
201 are elevated in the cloudinid interval. Mg/Ca ratios show the dominance of limestone in the Gaojiashan
202 Member, with higher Mg/Ca, Mn/Sr and Rb/Sr ratios found only in dolostones interbedded with siltstones in
203 the lower Gaojiashan Member. Sr/Ca ratios reveal a positive excursion in the upper section, mimicking the
204 $\delta^{13}\text{C}_{\text{carb}}$ anomaly, whereas Ce/Ce* ratios through the section remain constant at values near to 0.5, with the
205 exception of two samples at the top of the Gaojiashan Member with higher values.

206 207 DISCUSSION

208 Diagenesis

209 Confidence in our ability to interpret environmental changes associated with the paleontological
210 transitions in the Gaojiashan requires that we evaluate the degree of alteration of the limestone samples. Based
211 on the low Mn/Sr and Rb/Sr, and the smooth temporal trends in other geochemical indicators defined by high-
212 resolution sampling, the limestones appear to be especially well preserved (**Fig. 4**). However, insofar as

213 carbonates are susceptible to isotopic exchange with meteoric or hydrothermal fluids after burial, stable isotope
214 compositions of carbonate phases might reflect diagenetic overprints over depositional signatures. For example,
215 the lithification of marine carbonates associated with the flushing of meteoric fluids could cause coupled
216 depletions in both ^{13}C and ^{18}O , assuming the alkalinity was sourced from soil respiration (Knauth and Kennedy,
217 2009). Isotopic coupling in carbonates might also result from burial diagenesis (Derry, 2010; Bristow et al.,
218 2011) assuming hot fluid temperatures and alkalinity formed through anaerobic processes. In either case the
219 carbonates would be predictably recrystallized or contain appreciable amounts of neomorphic calcite. These
220 petrographic features are not observed in the fine-grained Gaojiashan limestones, which reveal a significant
221 positive $\delta^{13}\text{C}_{\text{carb}}$ excursion whereas $\delta^{18}\text{O}_{\text{carb}}$ values remain steady. A cross plot of the carbon and oxygen
222 isotope abundances in these samples reveals no positive correlation (**Fig. 5, lower panel**). Insofar as oxygen
223 isotopes would be more likely to be altered during water-rock interactions (Jacobsen and Kaufman, 1999), the
224 $\delta^{13}\text{C}_{\text{carb}}$ excursion recorded in the Gaojiashan Member is likely to reflect true secular changes in seawater
225 composition.

226
227 The degree of carbonate preservation may also be evaluated through the analyses of CAS abundances
228 and sulfur isotope compositions. Published studies have shown that CAS in marine carbonates may be affected
229 by secondary processes related to pyrite oxidation (Marenco et al., 2008), which could occur in the outcrop or
230 in the laboratory, or the addition of secondary atmospheric sulfate (SAS) to carbonates exposed in desert
231 environments (Peng et al., 2014). On the other hand, CAS studies of modern carbonate sediments where there
232 was active pore-water sulfate reduction indicate minimal alteration of bulk carbonate sulfur isotope
233 compositions (Lyons et al., 2004). While Gaojiashan CAS abundances are generally low (ranging from near 0
234 to 150 ppm), their $\delta^{34}\text{S}_{\text{CAS}}$ values are invariant at ca. +40‰ throughout the succession (**Fig. 5F,G**), suggesting
235 excellent preservation of primary signals (Gill et al., 2008). The sulfur isotope invariance, which is notably
236 consistent with $\delta^{34}\text{S}_{\text{sulfate}}$ analyses of bedded anhydrites (ca. +40‰) in equivalent terminal Ediacaran strata
237 from Oman (Fike and Grotzinger, 2008), supports the view that the Gaojiashan carbonates are exceptionally
238 well preserved and likely reflective of global seawater conditions.

239
240 To further evaluate diagenesis in the Gaojiashan Member samples we compared abundances of TOC
241 and pyrite against each other, as well as with their carbon and sulfur isotope compositions, respectively. In
242 neither case do we see a systematic relationship (**Fig. 5 lower panel**), although the two samples with the
243 highest TOC do have the lowest $\delta^{13}\text{C}$ signatures. TOC might change by either microbial (Borowski et al., 1996;
244 Jørgensen et al., 2004; Ries et al., 2009; Borowski et al., 2013) or thermochemical (Cai et al., 2001; Cai et al.,
245 2003; Cai et al., 2004) sulfate reduction after deposition, which could result in progressive ^{34}S -enrichment of
246 product sulfide preserved as pyrite. However, the sedimentary rocks have not been buried deeply enough to
247 drive the thermal reactions. Furthermore, we find no systematic relationship in TOC–TS or TOC– $\delta^{34}\text{S}$ cross
248 plots, suggesting that these secondary processes did not significantly impact the Gaojiashan samples.

249 ***Redox constraints for the Gaojiashan Member***

250 Multiple lines of evidence suggest that the paleontological transition in the middle Gaojiashan
251 Member is accompanied by strong ocean stratification (**Figs. 5, 6**). Support for this interpretation comes from
252 the negative excursion in the ^{13}C abundance of total organic carbon (TOC) – which mirrors the positive
253 $\delta^{13}\text{C}_{\text{carb}}$ excursion – resulting in the greatest degree of $\Delta\delta^{13}\text{C}$ in the *Cloudina* interval (**Fig. 5 A-C**). In light of
254 the abundance of microbial fabrics in both the Gaojiashan Member (Cai et al., 2010) and the Nama group
255 (Bouougri and Porada, 2007), these decoupled chemostratigraphic carbon isotope trends could plausibly have
256 resulted from organic matter derived from benthic microbial mats where anaerobic chemoautotrophs utilized
257 locally recycled ^{13}C -depleted DIC to form biomass with lower $\delta^{13}\text{C}_{\text{org}}$ values (Des Marais, 1990; Hayes, 1993;
258 Falkowski et al., 2008; Houghton et al., 2014). Alternatively, there may have been an enhanced flux of organic
259 matter derived from anoxygenic photoautotrophs such as green and purple sulfur bacteria that utilize H_2S as a
260 source of electrons during photosynthesis (Johnston et al., 2009). These photoautotrophs typically exist along
261 redox chemoclines and utilize respired CO_2 , which is typically depleted in ^{13}C relative to its atmospheric
262 equivalent (e.g. Brocks et al., 2005). In either case, the spread of anoxic/euxinic conditions across the platform
263 would have promoted organic matter burial (Hayes et al., 1983) and the positive $\delta^{13}\text{C}_{\text{carb}}$ excursion. High
264 abundances of organic S compounds, indicative of euxinic conditions, are also revealed by biomarker studies
265

266 of the Gaojiashan-equivalent Shibantan Member (Duda et al., 2014). Further evidence for the spread of anoxia
 267 associated with the decoupled $\delta^{13}\text{C}$ excursions is found in the profoundly negative $\delta^{238}\text{U}$ signatures of
 268 Gaojiashan limestones (Zhang et al., 2015), and our sulfur isotope measurements.

269
 270 Chemostratigraphic analyses of the Gaojiashan Member reveal a profound rise in $\delta^{34}\text{S}_{\text{pyrite}}$ values from
 271 as low as -30‰ in the lower half of the member to peak values near $+30\text{‰}$ between 35 and 40 m before falling
 272 rapidly to values averaging around $+10\text{‰}$ in the *Cloudina*-bearing beds (**Fig. 5G**). In contrast, the $\delta^{34}\text{S}$ of
 273 carbonate associated sulfate (CAS) remain steady at values of ca. $+40\text{‰}$ throughout the Gaojiashan Member.
 274 The calculated sulfur isotope contrasts ($\Delta\delta^{34}\text{S}$) range widely in the lower half of the succession, with a
 275 maximal value of 72‰ , but in the upper half $\Delta\delta^{34}\text{S}$ is relatively constant at ca. $30 - 35\text{‰}$ (**Fig. 5H**).
 276 Interpreting the environmental significance of these remarkable stratigraphic variations requires the
 277 recognition that the $\delta^{34}\text{S}_{\text{CAS}}$ and $\delta^{34}\text{S}_{\text{pyrite}}$ signatures are inherited from different parts of the depositional basin.
 278 Sulfate incorporation into primary carbonate sediments would occur within the water column, whereas pyrite
 279 would form either in euxinic bottom waters or within sediments. Considering this spatial separation, local
 280 sulfate availability could dictate the $\delta^{34}\text{S}$ isotopic difference between CAS and pyrite, particularly if pyrite is
 281 formed in non-bioturbated and microbially-sealed sediments where the water-sediment interface represents a
 282 significant diffusion barrier (Seilacher and Pflüger, 1994; Bottjer et al., 2000; Bouougri and Porada, 2007; Fike
 283 et al., 2008; Fike et al., 2009). While such a scenario might apply to discrete intervals within the Gaojiashan –
 284 including the *Shaanxilithes* and *Conotubus* zones (**Fig. 5H**) – other parts of the succession have measured
 285 $\Delta\delta^{34}\text{S}$ differences that are significantly larger. Furthermore, the constancy of the $\delta^{34}\text{S}_{\text{CAS}}$ values through the
 286 Gaojiashan suggests that the perturbation in the terminal Ediacaran sulfur cycle did not involve changes in the
 287 marine sulfate isotopic composition. Thus, the $\sim 60\text{‰}$ shift in $\delta^{34}\text{S}$ of pyrite from the lower to the upper
 288 Gaojiashan Member may require a change in biologically-induced fractionations involving both the reductive
 289 and oxidative paths of the sulfur cycle (i.e. bacterial S disproportionation, or BSD) (Canfield and Thamdrup,
 290 1994), or microbial sulfate reduction (MSR) with very low sulfate reduction rates (SRR) (Canfield et al., 2010;
 291 Leavitt et al., 2013; Wu and Farquhar, 2013; Wing and Halevy, 2014).

292
 293 In the case of BSD, sulfur is recycled via both reductive and oxidative pathways. On the reductive side,
 294 the magnitude of kinetic sulfur isotope fractionation (ϵ_{SR}) has been observed to correlate directly with
 295 extracellular sulfate concentrations. Experiments from pure cultures of sulfate reducers indicate maximal
 296 fractionation of 66‰ at sulfate concentrations similar to modern seawater at 28 mM (Sim et al., 2011), while
 297 ϵ_{SR} may be suppressed at very low sulfate abundances ($<200 \mu\text{M}$) (Habicht et al., 2002). On the oxidative side,
 298 the sulfide produced through MSR is typically re-oxidized to elemental sulfur, which is subsequently
 299 disproportionated to sulfate and sulfide, by coupling with the reduction of O_2 , NO_3^- , or iron and manganese
 300 compounds (Canfield and Thamdrup, 1994). Disproportionation reactions thus can significantly augment the
 301 fractionations induced during MSR, resulting in isotopic contrasts between reactant sulfate and product sulfide
 302 of greater than 70‰ (**Fig. 5H**).

303 Alternatively, very low rates of MSR may also lead to large fractionations. Recent studies of
 304 lacustrine euxinic systems indicate that $>70\text{‰}$ fractionations are achievable by both isolated and natural
 305 populations of sulfate reducers (Canfield et al., 2010; Gomes and Hurtgen, 2015). Furthermore,
 306 environmentally controlled experiments suggest that MSR-related fractionation could be strain specific
 307 (Bradley et al., 2016), or related to sulfate reduction rates that are dependent on the availability of organic
 308 substrates as electron donors (Canfield et al., 2010; Leavitt et al., 2013; Leavitt, 2014; Wing and Halevy, 2014;
 309 Gomes and Hurtgen, 2015). In this case the magnitude of fractionation is inversely correlated with the rate of
 310 sulfate reduction (e.g., Xiao et al., 2010). With these constraints in mind, the $\Delta\delta^{34}\text{S} >70\text{‰}$ in the lower interval
 311 of the Gaojiashan may reflect BSD coupled with MSR, or result solely from MSR with very low SRR. The
 312 former scenario is consistent with sulfide oxidation occurring along a chemocline above euxinic deep waters
 313 (**Fig. 6A**), which is our preferred interpretation for this marginal marine basin.

314 Based on systematic studies of modern environments and Phanerozoic shales (Bernier and Raiswell,
 315 1983; Bernier and Raiswell, 1984), very low C/S ratios might indicate euxinic marine conditions. Although the
 316 C/S proxy is not well calibrated for carbonates, the preponderance of Gaojiashan limestones with values <1
 317 (**Fig. 5 lower panel**) suggests the possibility of euxinic conditions in the depositional basin. This view is

318 consistent with the high Ce/Ce* values in the upper Gaojiashan samples (**Fig. 4H**), although carbonates have
 319 notoriously low REE abundances and should thus be interpreted with caution.

320

321 *Global indicators of dynamic redox conditions*

322 Chemostratigraphic comparison of terminal Ediacaran successions in South China, Oman, and
 323 Namibia reveal both similarities and differences, suggesting local overprint of global signals in some basins
 324 (e.g., Loyd et al., 2013; Wood et al., 2015) (**Fig. 7**). Global conditions appear to be reflected in the similarity in
 325 the magnitude and direction of isotope trends in South China and Oman. For example, in the uppermost Buah
 326 Formation of Oman, paired CAS-pyrite measurements reveal large magnitude sulfur isotope fractionations
 327 (with maximal $\Delta\delta^{34}\text{S} \sim 50\text{‰}$) prior to the first occurrence of *Cloudina* (Conway Morris et al., 1990), just as we
 328 document for the lower Gaojiashan Member. The large fractionation seen in two basins suggests the
 329 dominance of sulfur disproportionation reactions (Fike et al., 2006; Fike and Grotzinger, 2008), which is
 330 supported by a recent multiple sulfur isotope study indicating enhanced sulfide re-oxidation in the uppermost
 331 Buah (Wu et al., 2015). In this interval disproportionation reactions likely dominated over MSR insofar as the
 332 latter would be discouraged if there was active photoautotrophic sulfide oxidation (**Fig. 6A**) (Habicht and
 333 Canfield, 2001). Stratigraphically higher in the Ara Formation where *Cloudina* occurs in carbonates
 334 interbedded with evaporites, the $\delta^{34}\text{S}$ compositions of pyrite and CAS are notably invariant with a smaller
 335 magnitude of fractionation (ca. 30‰) (Fike and Grotzinger, 2008), again exactly matching our observations
 336 from the upper Gaojiashan. The ^{34}S enrichments in pyrite and the smaller sulfur isotope differences between
 337 reduced and oxidized phases are best explained by high rate of MSR, which we view as the dominant sulfur
 338 metabolism associated with the spread of anoxic bottom waters (**Figs. 6B-C**). In sum, the correlated
 339 observations from South China and Oman indicate a global environmental control on biological sulfur
 340 fractionations.

341

342 To the contrary, chemostratigraphic data from *Cloudina*-bearing strata of the Nama Group in southern
 343 Namibia provide a completely different pattern of ^{34}S enrichments and fractionation. In this case strongly
 344 positive $\delta^{34}\text{S}_{\text{pyrite}}$ values are most-often paired with anomalously low and scattered $\delta^{34}\text{S}_{\text{CAS}}$ values, resulting in
 345 inversely fractionated $\Delta\delta^{34}\text{S}$ values (Ries et al., 2009). Stratigraphically coherent CAS results are only seen in
 346 the Omkyk Member where there is a positive $\delta^{34}\text{S}_{\text{CAS}}$ shift from ca. +10 up to +40‰, which is similar to the
 347 trend encompassing the transition to the *Cloudina*-bearing beds in Oman and South China (**Fig. 7**), and in the
 348 uppermost Spitzkopf Member below the Ediacaran-Cambrian boundary. In this case, however, the CAS sulfur
 349 isotope compositions are depleted in ^{34}S by $\sim 20\text{‰}$ relative to equivalent upper Ara strata in Oman. Based on
 350 our experience with quantitative preparation techniques, it would appear that the Namibian CAS samples were
 351 not adequately leached of non-CAS components. In contrast to our extensive efforts to remove the non-
 352 carbonate fraction (see Methods), Ries et al. (2009) leached the CAS powders with Milli-Q water only once,
 353 and this is unlikely to have removed sulfate on mineral surfaces formed through pyrite oxidation (Marengo et
 354 al., 2008) or secondary atmospheric sulfate (SAS) (Peng et al., 2014). The presence of these contaminants
 355 would cause $\delta^{34}\text{S}_{\text{CAS}}$ values to be more negative and hence would not reflect depositional signatures (Wotte et
 356 al., 2012). In our view the inversely fractionated sulfur isotopes from this succession should be interpreted
 357 with caution, although they do highlight the potential redox differences between equivalent terminal Ediacaran
 358 basins.

359

360 In addition, there are notable contrasts in carbon isotope anomalies among the terminal Ediacaran
 361 successions in South China, Oman, and Namibia. Chemostratigraphic data from the Dengying Formation
 362 suggests the possibility of three separate positive excursions (**Fig. 1**), with their different stratigraphic expanses
 363 likely associated with varying sediment accumulation rates. In contrast, there is significant $\delta^{13}\text{C}_{\text{carb}}$ variability
 364 in the evaporite-rich succession from Oman (Fike and Grotzinger, 2008; Wu et al., 2015), including negative
 365 anomalies within the cloudinid interval interspersed with at least two positive excursions (**Fig. 7**). In Namibia
 366 there is only one post-Shuram positive $\delta^{13}\text{C}_{\text{carb}}$ excursion followed by a long plateau of moderately positive (ca.
 367 +1 to +3‰) values leading up to the Ediacaran-Cambrian boundary (**Fig. 7**) (Ries et al., 2009). Other terminal
 368 Ediacaran successions, including those in northern India, also reveal significant differences in carbon isotope
 369 stratigraphic profiles (Kaufman et al., 2006). Taken together, the inter-basinal variations in carbon and sulfur
 370 isotope compositions likely reflect redox differences in the depositional environments of the various basins. If

371 correct, the Ediacaran experiment in animal life must have been spread across a dynamic environmental
372 landscape, which may help to explain the distribution of geographically unique assemblages (Narbonne et al.,
373 2014)..

374

375 **Enhanced alkalinity in the terminal Ediacaran ocean**

376 Compared with carbonates in the underlying Doushantuo Formation (e.g., McFadden et al., 2008), the
377 generally lower TOC contents of the Gaojiashan limestones are notable, and may reflect either depositional or
378 early diagenetic processes. For example, pervasive water column or sediment recycling of organic matter may
379 have decreased original organic carbon contents in sediments. In addition, the anaerobic conversion of simple
380 organic compounds to alkalinity could have resulted in the formation of ubiquitous authigenic carbonates
381 (Higgins et al., 2009; Schrag et al., 2013). Driven by iron or sulfate reduction of available organic substrates,
382 the addition of authigenic carbonate to the sediments would, however, have resulted in a negative (rather than a
383 positive) carbon isotope excursion. Alternatively, the generally lower TOC values may reflect significant
384 dilution by abundant carbonate formed from highly alkaline seawater. In this case the source of the alkalinity
385 was more likely to be from terrestrial weathering, as indicated by the significant rise in $^{87}\text{Sr}/^{86}\text{Sr}$ in the terminal
386 Ediacaran Period (Kaufman et al., 1993; Kaufman et al., 1997; Halverson et al., 2007; Sawaki et al., 2010).
387 The abundance and carbon isotopic composition of river-derived alkalinity in the Ediacaran Period was likely
388 to be high variable, depending on the differential weathering of bedrock lithologies (e.g. carbonates vs.
389 silicates). In the absence of land plants or extensive microbial surfaces, terrestrial sources of alkalinity need not
390 have been significantly depleted in ^{13}C . Elevated seawater alkalinity at this time is consistent with the
391 presence of aragonite crystal fans preserved in the time-equivalent Nama Group in Namibia (Grotzinger, 2000;
392 Grotzinger et al., 2005; Hall et al., 2013), which records a singular Ediacaran positive $\delta^{13}\text{C}_{\text{carb}}$ excursion
393 (Kaufman et al., 1991; Saylor et al., 1998). Overall high alkalinity is consistent with the dominance of
394 carbonate in terminal Ediacaran successions worldwide, and may explain the extremely high accumulation rate
395 estimated for the Dengying (i.e. >650 m in ~10 million years), as compared with the underlying Doushantuo (i.e.
396 <200 m in ~84 million years) (**Fig. 1**).

397

398 Enhanced terminal Ediacaran alkalinity may also be interpreted from our elemental results from the
399 Gaojiashan, as well as inter- and intra-basinal equivalents. In particular the [Sr] and Sr/Ca data measured from
400 the Gaojiashan limestones reveal positive excursions in step with the positive $\delta^{13}\text{C}_{\text{carb}}$ anomaly (**Fig. 8**). A
401 similar [Sr] excursion coincident with peak $\delta^{13}\text{C}$ compositions of carbonates is noted in the Gaojiashan-
402 equivalent Shibantan Member in the Yangtze Gorges area (Sawaki et al., 2010). A rise in [Sr] is also noted in
403 the broadly equivalent Nama Group in southern Namibia (Ries et al., 2009) although this geochemical
404 anomaly post-dates the first appearance of *Cloudina* and the singular positive $\delta^{13}\text{C}$ excursion in the thick
405 sedimentary succession. The apparent [Sr] rise in Namibia is potentially complicated by the admixture of
406 siliciclastics within the carbonates by using aqua regia acid, which would attack both carbonate and siliclastic
407 components in the dissolution procedure of Ries et al. (2009). Nonetheless, all sections show positive [Sr]
408 excursion in *Cloudina*-bearing intervals. For the Gaojiashan, the invariantly low Mg/Ca values of samples
409 suggests that dolomitization played no role in the elemental excursion (**Fig. 4**).

410

411 Given that the dominant source of Sr in the ocean is from the chemical weathering of the continental
412 crust (Shields, 2007; McArthur et al., 2012), including both silicates and carbonates, enhanced [Sr] and Sr/Ca
413 values supports the view that terrestrial weathering and the delivery of alkalinity through riverine inputs
414 buffered shallow ocean basins to variable degrees in the terminal Ediacaran Period. The weathering of Ca
415 silicate minerals can be simply represented by the overall reaction (Berner, 2004): $\text{CO}_2 + \text{CaSiO}_3 \rightarrow \text{CaCO}_3 +$
416 SiO_2 , which over geological time scales sequesters atmospheric CO_2 into carbonate minerals by liberating Ca^{2+}
417 and HCO_3^- ions that are then carried to seawater by rivers. While terrestrial carbonate weathering is not a
418 geological sink for atmospheric CO_2 , the resultant flux of alkalinity does effect overall carbonate saturation
419 state (e.g., Kump et al., 1999; Hoffman and Schrag, 2002; Higgins and Schrag, 2003).

420

421 Associated with higher alkalinity and higher carbonate saturation, it should be noted that Sr/Ca may
422 also be controlled by precipitation rate. For biogenic carbonates (e.g., coccoliths) the ratio of Sr to Ca has been
423 widely used as a productivity proxy (e.g., Stoll and Schrag, 2001; Stoll and Bains, 2003) insofar as there is a

424 strong link between Sr/Ca, export production, and calcification rate (Stoll and Schrag, 2001). Similarly,
425 laboratory experiments reveal that rapid precipitation rates induce greater Sr partitioning into abiotic calcite
426 (Lorens, 1981; Tesoriero and Pankow, 1996; Tang et al., 2008; DePaolo, 2011). Thus, the precipitation rate of
427 carbonate in either *Cloudina* or in inorganic micrite may have additionally influenced the Sr/Ca ratios of the
428 Gaojiashan limestones.

429
430 Variable partitioning of strontium by aragonite and calcite relative to seawater may also have resulted
431 in the observed variations of Sr/Ca in the Gaojiashan samples. Strontium has a crystal ionic radius larger than
432 that of Ca^{2+} and thus prefers the more open octahedral crystal structure of aragonite over the smaller hexagonal
433 structure of calcite (Wray and Daniels, 1957; Lorens, 1981). Thus on one hand, the enhancement in [Sr] in the
434 Gaojiashan limestones and their equivalents might reflect a secular change from calcite to aragonite-dominated
435 seas (e.g., Stanley and Hardie, 1998), although the short stratigraphic interval represented by the Gaojiashan
436 Member would seem to preclude this possibility. On the other, the [Sr] excursion might result from post-
437 depositional diagenesis, through which Sr was preferentially flushed from specific horizons in the Gaojiashan
438 Member during neomorphic aragonite-to-calcite transformations (Katz et al., 1972). However, given the
439 excellent petrographic and oxygen isotopic preservation of the samples, the absence of dolomitization, and
440 smooth carbon and sulfur isotope trends, we attribute the elevated Sr/Ca ratios in the *Cloudina* interval to 1)
441 elevated Sr flux from the continents to contemporaneous seawater, and/or 2) enhanced precipitation rate in a
442 carbonate over saturated ocean.

443
444 Emerging support for weathering-induced high alkalinity in terminal Ediacaran seawater may come
445 from Ca isotope ($\delta^{44}\text{Ca}$) measurements of the Gaojiashan equivalent Shibantan Member. In this unit a sharp
446 negative $\delta^{44}\text{Ca}$ excursion (down to 0.3‰) has been interpreted to reflect anomalously high seawater Ca
447 concentrations. The Ca isotope system has been used to make inferences regarding the seawater Ca cycle in
448 deep time (DePaolo, 2004; Nielsen et al., 2011), with a special emphasis on perturbations of $\delta^{44}\text{Ca}$ during
449 chemical weathering events. Enhanced chemical weathering in the Cenozoic (between 40 and 10 Ma), for
450 example, has been interpreted based on the presence of negative $\delta^{44}\text{Ca}$ excursions (De La Rocha and DePaolo,
451 2000; DePaolo, 2004). Similarly, a Cretaceous negative $\delta^{44}\text{Ca}$ excursion believed to be related to enhanced
452 weathering (Blättler et al., 2011) coincides with a positive $\delta^{13}\text{C}$ anomaly and an Oceanic Anoxic Event, all of
453 which match our observations of the Gaojiashan.

454
455 In concert, the sedimentological and geochemical observations of *Cloudina*-bearing strata in South
456 China and elsewhere suggest that the terminal Ediacaran ocean was highly alkaline. If correct, high
457 concentrations of Ca and alkalinity well may have enabled the earliest example of calcareous
458 biomineralization by animals.

459
460 ***Environmental context of pyritization and biomineralization***

461 Based on our chemostratigraphic observations of the Gaojiashan Member, the first appearance of
462 *Cloudina* coincided with the development of anoxic and episodically euxinic conditions across the shelf
463 environment. It is likely, however, that this first biomineralized metazoan lived in the oxidized shallower water
464 column and was swept into deeper anoxic settings by storm events (Cai et al., 2010) (**Fig. 5B**). In contrast, our
465 geochemical results suggest that *Conotubus* and other soft-bodied Ediacara biotas thrived at a time of more
466 generally oxidizing conditions within the water column prior to the peak of the $\delta^{13}\text{C}$ excursion. In the
467 equivalent Shibantan Member in the Yangtze Gorges region, the soft-bodied organisms are preserved in
468 subtidal environments and are closely associated with abundant bilaterian burrows, suggesting moderate levels
469 of bioturbation (Chen et al., 2013; Chen et al., 2014a; Meyer et al., 2014). It is notable that *Conotubus* and
470 many Ediacara remains, as well as the microbial surfaces, were preserved in these environments through
471 pyritization (Gehling, 1999; Schiffbauer et al., 2014). In the “death mask” model, pyritization of a
472 decomposing metazoan would stabilize its surface and allow the external form of the organism to be imprinted
473 with exquisite detail; in the case of the Gaojiashan Member, similar pyritization process may have also molded
474 *Conotubus* tubes from inside. Based on *in-situ* SIMS $\delta^{34}\text{S}$ analyses, it has been proposed that pyritization of
475 *Conotubus* was fueled by the degradation of labile organic tissues through MSR (Schiffbauer et al., 2014) near
476 the sediment-water interface. Consistent with pyritization as a widespread fossilization pathway, many soft-

477 bodied fossils and associated microbial surfaces in the Gaojiashan and Shibantan exposures are coated with
478 iron oxides and jarosite (an iron-bearing sulfate mineral) that are the oxidative weathering products of early
479 diagenetic pyrite (Hall et al., 2013).

480
481 The balance between the ecological pressures and physiological responses that resulted in the
482 biomineralization of *Cloudina* are nicely viewed from the Gaojiashan Member and its equivalents in South
483 China. On one hand, Hua et al. (2003) highlighted the large number of drill holes on *Cloudina* shells in the
484 Dengying Formation and hypothesized that predation tipped the balance towards calcification as a means of
485 protection. On the other, our results emphasize clear temporal changes in seawater chemistry that are
486 associated with this evolutionary milestone. We interpret the geochemical trends to reflect enhanced terminal
487 Ediacaran chemical weathering that introduced nutrients, which drove primary productivity, the spread of
488 anoxia, and higher rates of organic carbon burial. Chemical weathering would also have delivered alkalinity
489 and cations including calcium to seawater, promoting rapid carbonate accumulation in shallow marine settings.
490 In addition, sulfate delivery would have further stimulated MSR, which would provide an additional source of
491 seawater alkalinity depending on the extent of water column anoxia. Biomineralization could then have been a
492 means to remove excess calcium from the newly-developed circulatory systems of evolving metazoans
493 (Simkiss, 1977) in the context of higher overall seawater alkalinity in the terminal Ediacaran Period
494 (Grotzinger et al., 2005).

495
496 Studies of Phanerozoic biomineralization further highlight the effect of seawater chemistry on
497 calcification. Although the secretion of biominerals often occurs in internal environments isolated from
498 seawater (Weiner and Dove, 2003), seawater chemistry could indirectly determine skeletal mineralogy by
499 affecting the physiological costs of biomineralization (Knoll, 2003a), thus resulting in distinct patterns of
500 skeleton evolution through Earth history. Indeed, extensive compilation of non-skeletal carbonates and
501 hypercalcifying animals in the Phanerozoic reveals that the Mg/Ca and [Ca] of seawater during periods of
502 aragonite or calcite-dominated seas had a strong influence on skeletal mineralogy (Stanley and Hardie, 1998;
503 Stanley, 2006; Porter, 2010). High-Mg calcite and aragonite shells appear to have dominated under aragonite
504 seas, while shells composed of low-Mg calcite dominated under calcite seas. This pattern is also seen in the
505 Cambrian Period when the first massive biodiversification of skeletal animals occurred (Porter, 2007). In the
506 terminal Ediacaran, the widespread appearance of seafloor aragonite fans (Grotzinger, 2000; Grotzinger et al.,
507 2005; Hall et al., 2013) and rapid accumulation of carbonates supports the aragonite sea hypothesis, and is
508 consistent with the inferred high-Mg calcite mineralogy of *Cloudina* (Grant, 1990; Zhuravlev et al., 2012).

509 ***Temporal growth in the oceanic sulfate reservoir***

511 Projected to the world stage and viewed through the long lens of Earth history, paired sulfur isotope
512 data in the terminal Ediacaran Gaojiashan Member stand out among the highest $\delta^{34}\text{S}_{\text{sulfate}}$ values (up to +40‰)
513 and largest S isotope fractionations (ca. +70‰) for the whole Precambrian (Fig. 7). This pattern may reflect a
514 strongly stratified ocean (e.g., Jiang et al., 2007; Shen et al., 2008; Li et al., 2010; Shen et al., 2010; Shen et al.,
515 2011), which would be particularly meaningful during the terminal Ediacaran when a putative atmospheric and
516 oceanic oxygenation event occurred (i.e. Neoproterozoic Oxygenation Event, or NOE) (Kaufman et al., 2007;
517 Shields-Zhou and Och, 2011; Och and Shields-Zhou, 2012; Lyons et al., 2014; Liu et al., 2016).

518
519 Growth of the Ediacaran sulfate pool has been hypothesized to be associated with the NOE (Fike et al.,
520 2006), which occurred in the aftermath of the Marinoan ice age (ca. 635 Ma). A recent quantitative model
521 analysis based on sulfur isotope trends through the Ediacaran Period suggests that oceanic $[\text{SO}_4^{2-}]$ was low (<5
522 mM) in the aftermath of the Marinoan glaciation, but then rose (>5 mM) across the Ediacaran-Cambrian
523 boundary (Algeo et al., 2015). The inflection of sulfate concentrations may well have coincided with the
524 middle Ediacaran Shuram Excursion, a profound negative carbon isotope anomaly (Grotzinger et al., 2011)
525 recognized in multiple sections across the globe including Shuram Formation in Oman and the upper
526 Doushantuo Formation of South China (Cui et al., 2015). Sulfur isotope profiles in both regions reveal a
527 parallel decrease in both $\delta^{34}\text{S}_{\text{pyrite}}$ and $\delta^{34}\text{S}_{\text{CAS}}$ (Cui et al., 2015) that likely reflect the significant growth of the
528 Ediacaran sulfate pool (Fike et al., 2006; Halverson and Hurtgen, 2007; Kaufman et al., 2007; McFadden et al.,
529 2008). Consistent with this view, pseudomorphs of gypsum are found in the Shuram equivalent Wonoka

530 Formation in South Australia (Calver, 2000) and in the Doushantuo (Lu et al., 2013). Above the Shuram, an
531 increasing number of bedded evaporite horizons have been discovered in terminal Ediacaran successions,
532 including the Dengying (Siegmund and Erdtmann, 1994; Duda et al., 2015; and this study) and Ara formations
533 (Fike and Grotzinger, 2010), as well as the Hanseran Evaporite Group in northwestern India (Strauss et al.,
534 2001; Mazumdar and Strauss, 2006).

535
536 The trigger for the terminal Ediacaran rise in seawater sulfate remains a question of considerable
537 debate. It has been proposed by Canfield and Farquhar (2009) that the emergence of bioturbation near the
538 Ediacaran-Cambrian boundary was the proximate cause insofar as sediment mixing would result in enhanced
539 sulfide oxidation and recycling (Bottjer et al., 2000; Meysman et al., 2006; Rogov et al., 2012; Chen et al.,
540 2014b). While bioturbation clearly modified ecosystems in the Fortunian Stage of the Cambrian Period
541 (Bottjer et al., 2000; Meysman et al., 2006), there is little support for deep penetration by animals into
542 sedimentary layers dominated by microbial mats in the Shuram or terminal Ediacaran intervals (e.g. Carbone
543 and Narbonne, 2014; Meyer et al., 2014). Moreover, recent investigations have demonstrated that the mixing
544 of sediments on marine shelves remained limited until at least the late Silurian, 120 million years after the
545 Precambrian–Cambrian transition (Tarhan and Droser, 2014; Gingras and Konhauser, 2015; Tarhan et al.,
546 2015).

547
548 Alternatively, it is likely that the increase in Ediacaran sulfate concentrations was driven by enhanced
549 oxidative weathering of pyrite in continental and oceanic sediments exposed by sea level regression (Kaufman
550 et al., 2007; Wang et al., 2016). This scenario is consistent with the profound increase in seawater $^{87}\text{Sr}/^{86}\text{Sr}$
551 (from ca. 0.7080 to 0.7090) recorded globally in carbonates deposited during the Shuram Excursion (Burns et
552 al., 1994; Calver, 2000; Melezhik et al., 2009; Sawaki et al., 2010; Cui et al., 2015). The Sr isotope shift most
553 likely accompanied enhanced silicate weathering (Kaufman et al., 1993; Halverson et al., 2007; Cui et al.,
554 2015), which led to an increase in the delivery of nutrient and sulfate to the oceans. By stimulating
555 photosynthesis, these continental fluxes would on the one hand result in the oxidation of surface environments,
556 while on the other, the remineralization of organic matter along marginal marine settings would simultaneously
557 resulted in the expansion of oxygen minimum zones (OMZs). Oceanic redox stratification would have
558 simultaneously stimulated the oxidative side of the sulfur cycle through widespread sulfur disproportionation
559 along chemoclines (Canfield and Thamdrup, 1994; Fike et al., 2006; Wu et al., 2015) and the reductive side
560 through microbial sulfate reduction within the anoxic plumes. Both microbial processes would have delivered
561 ^{32}S sulfur as pyrite into the sediments and thereby drove oceanic sulfate compositions to positive $\delta^{34}\text{S}$ extremes
562 (Canfield, 2004).

563 564 **CONCLUSIONS**

565 Terminal Ediacaran strata of the Gaojiashan Member preserve a record of dynamic exogenic carbon
566 and sulfur cycling ultimately driven by tectonic forces, a rise in atmospheric oxygen, and enhanced chemical
567 weathering of the exposed continents. Data from South China suggest that the attendant flux of nutrients and
568 alkalinity increased oceanic productivity and carbonate saturation state, resulting in a redox stratified ocean
569 basin where animals evolved to form calcareous shells, through the combined ecological pressure of predation
570 and the environmental pressure of high carbonate saturation, for the first time in Earth history.

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 583 manuscript.

584

585 **SUPPORTING INFORMATION**

586

587 **Table S1:** U-Pb geochronologic analyses of two detrital zircon samples in the lower Gaojiashan Member.

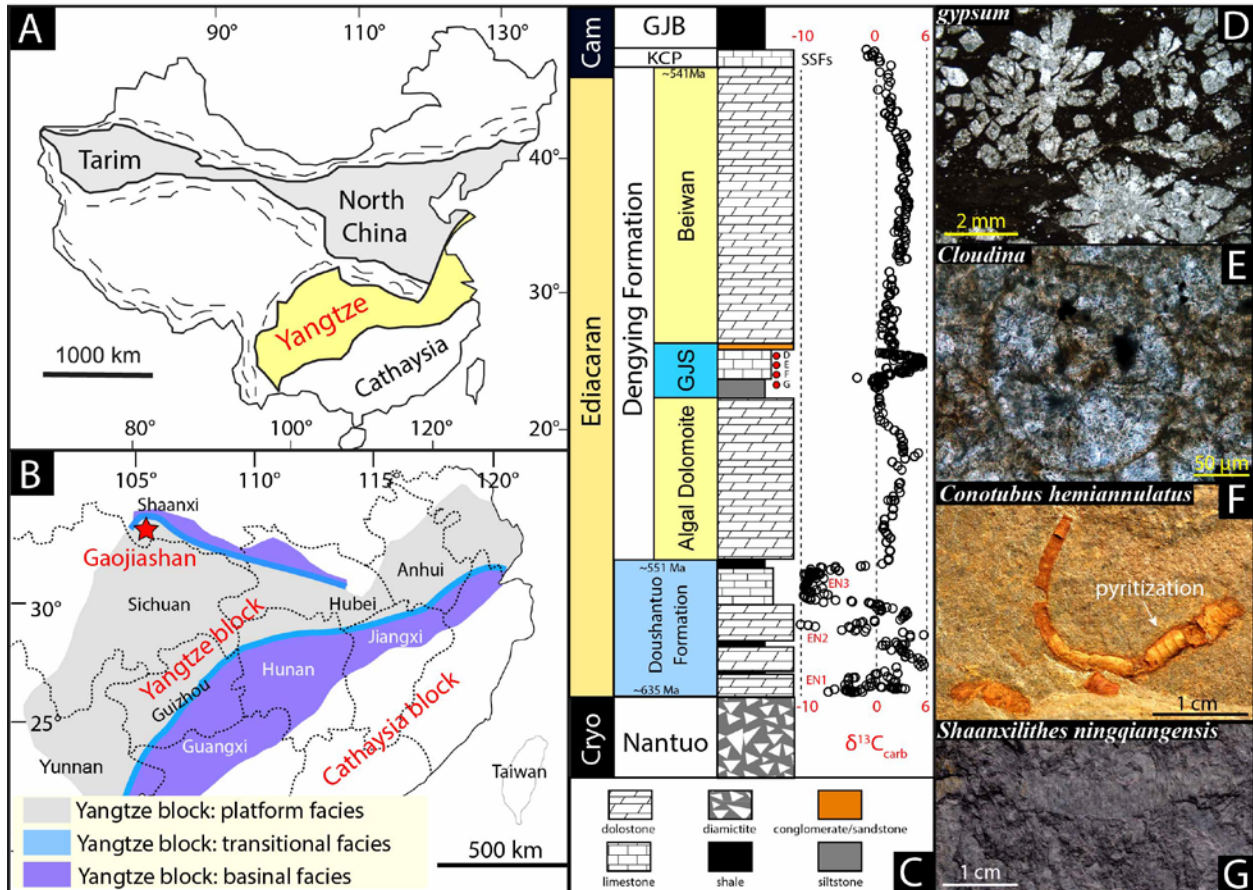
588 **Table S2:** Chemostratigraphic data of carbonate C isotopes of the Doushantuo and Dengying formations
 589 plotted in Figure 1.

590 **Table S3:** Chemostratigraphic data of C, O, and S isotopes of the Gaojiashan Member.

591 **Table S4:** Chemostratigraphic data of major and trace element concentrations of the Gaojiashan Member.

592

593 **FIGURES CAPTIONS**



594

595 **Figure 1.** (A) Tectonic framework of China, with the Yangtze Craton highlighted in yellow. (B) Ediacaran

596 depositional environments on the Yangtze Craton (Jiang et al., 2011). (C) $\delta^{13}\text{C}_{\text{carb}}$ record of the Dengying

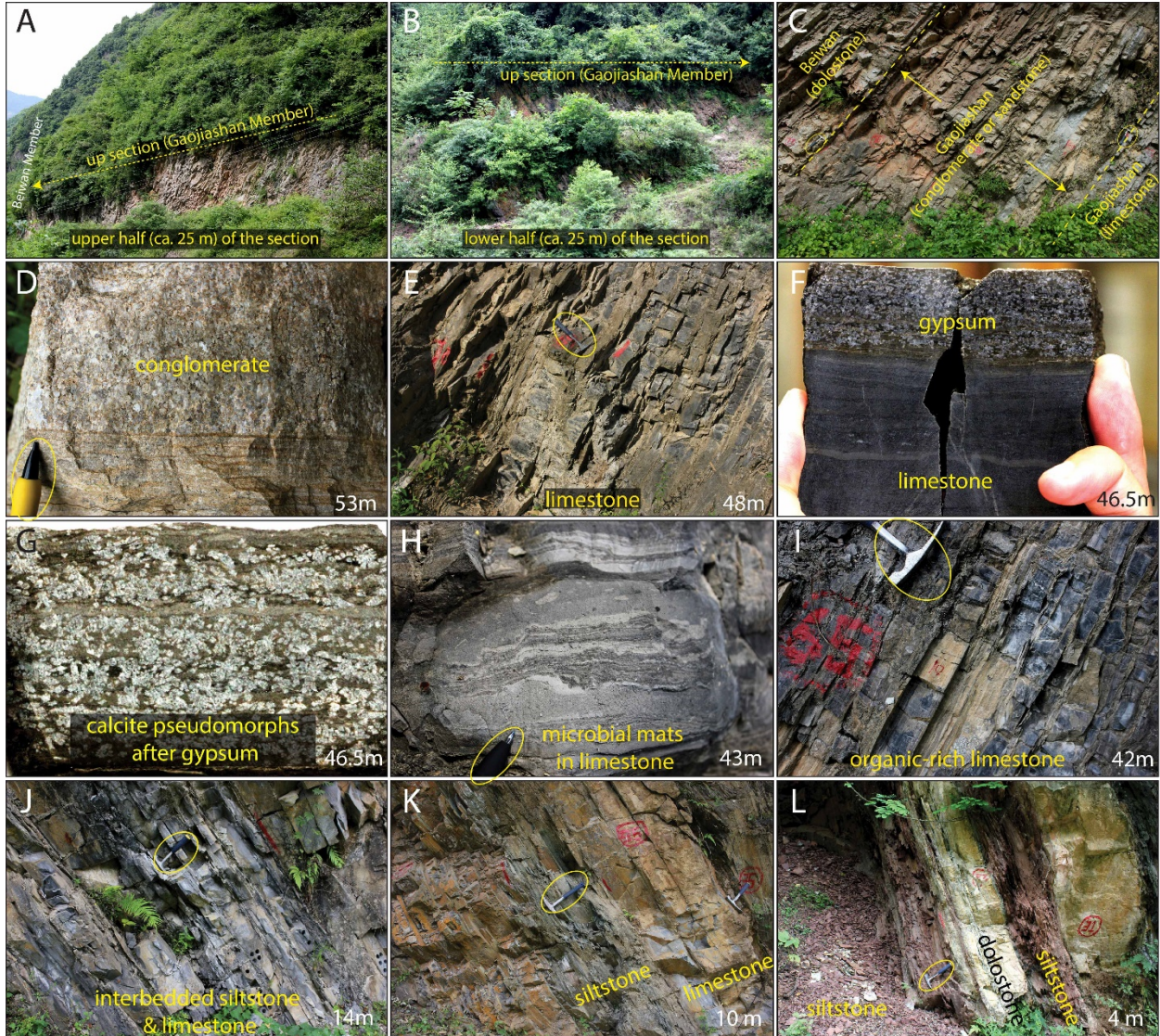
597 Formation in the Gaojiashan section. GJS = Gaojiashan, KCP = Kuanchuanpu, GJB = Guojiaba; $\delta^{13}\text{C}_{\text{carb}}$ data

598 for Doushantuo Formation are from McFadden et al. (2008). (D) Calcite pseudomorph after gypsum ca. 46.5 m

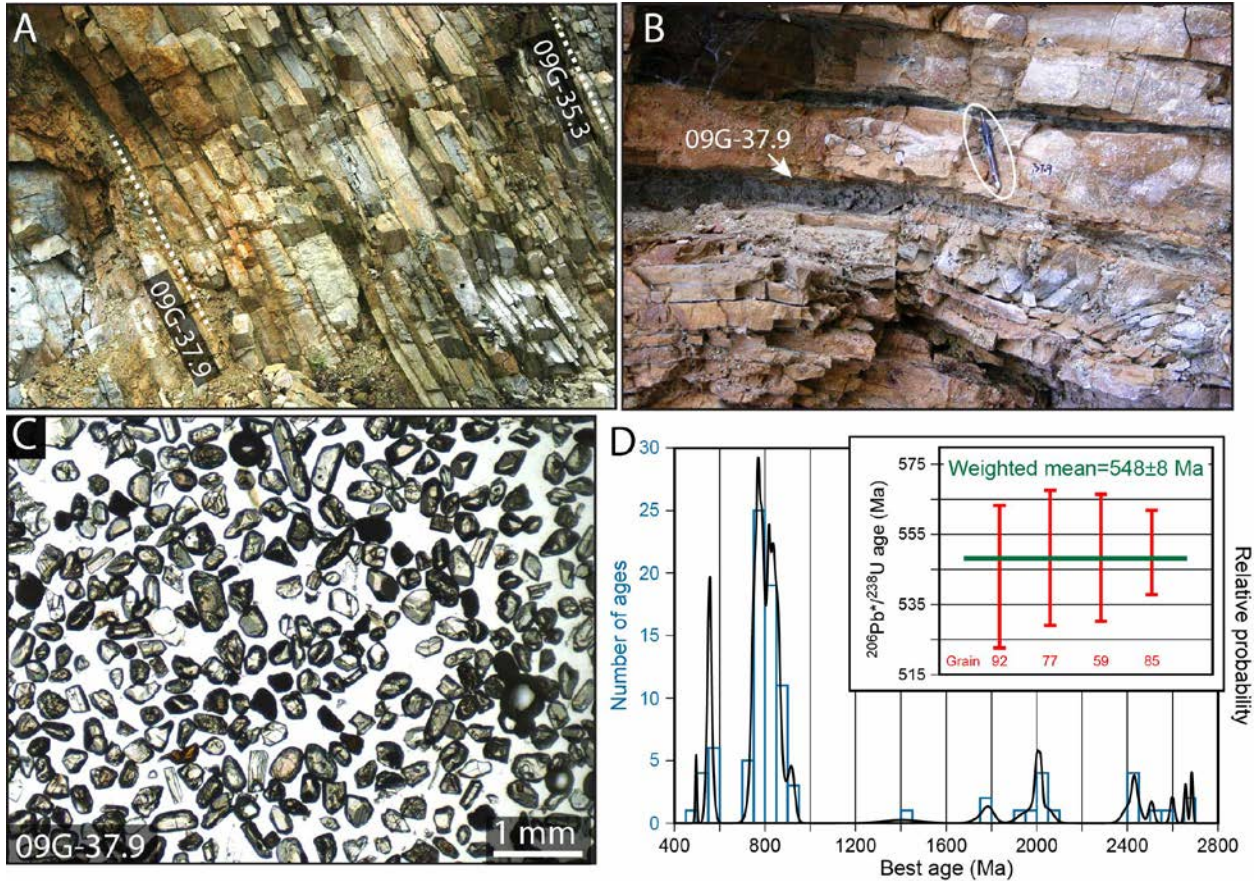
599 above the base of the Gaojiashan Member. (E) *Cloudina* in the Gaojiashan Member ca. 42 m above the base.

600 (F) Pyritized tubular fossil *Conotubus hemiannulatus* (Cai et al., 2011). (G) Enigmatic body fossil

601 *Shaanxilithes ningqiangensis* (Meyer et al., 2012).

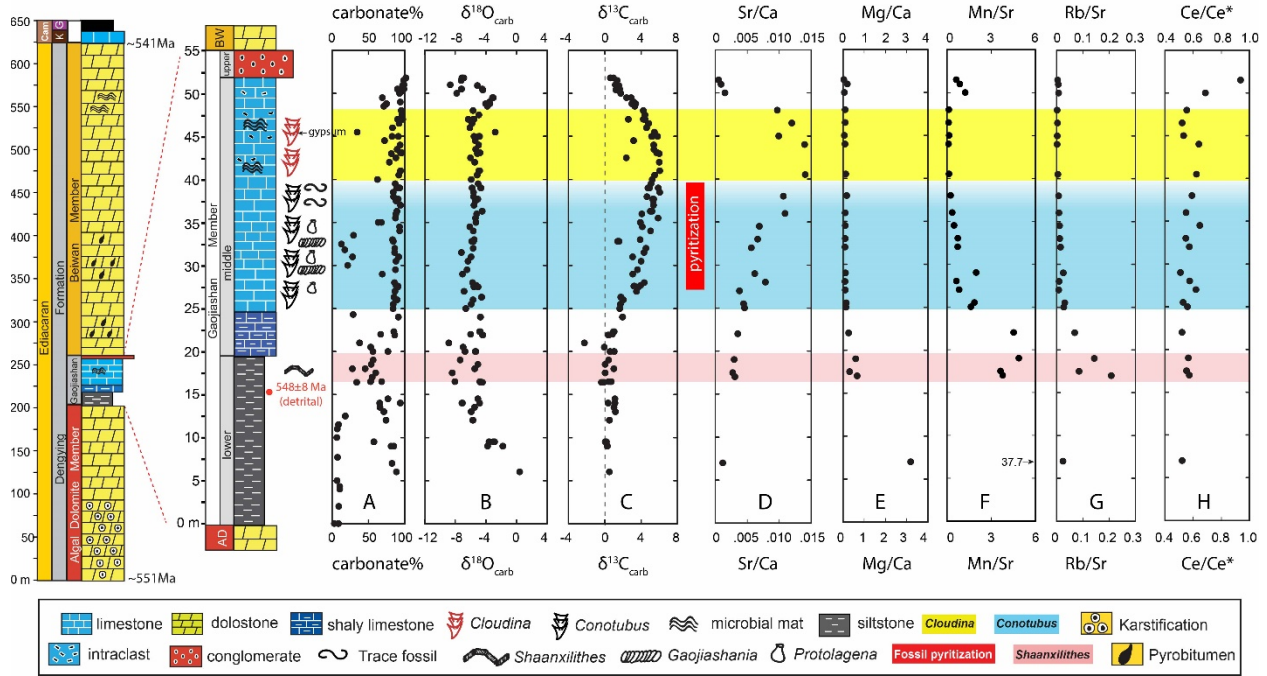


602
 603 **Figure 2:** Field photographs of the Gaojiashan Member. Hammers or pencils for scale and meterages above
 604 the base of the unit. (A, B) An overview of the upper and lower section (each ca. 25 meters in thickness). (C)
 605 Boundary between the Beiwan and Gaojiashan members. (D) Conglomerate in the uppermost interval. (E)
 606 Bedded limestone at 48 m. (F, G) Gypsum pseudomorphs at 46.5 m where crystals have been dissolved and
 607 replaced by calcite. (H, I) Organic-rich limestones with abundant microbial mats corresponding to the peak of
 608 the carbon isotope excursion at ca. 43 m. (J) Interbedded siltstone and limestone at 14 m. (K, L) Siltstones in
 609 the lower Gaojiashan Member.

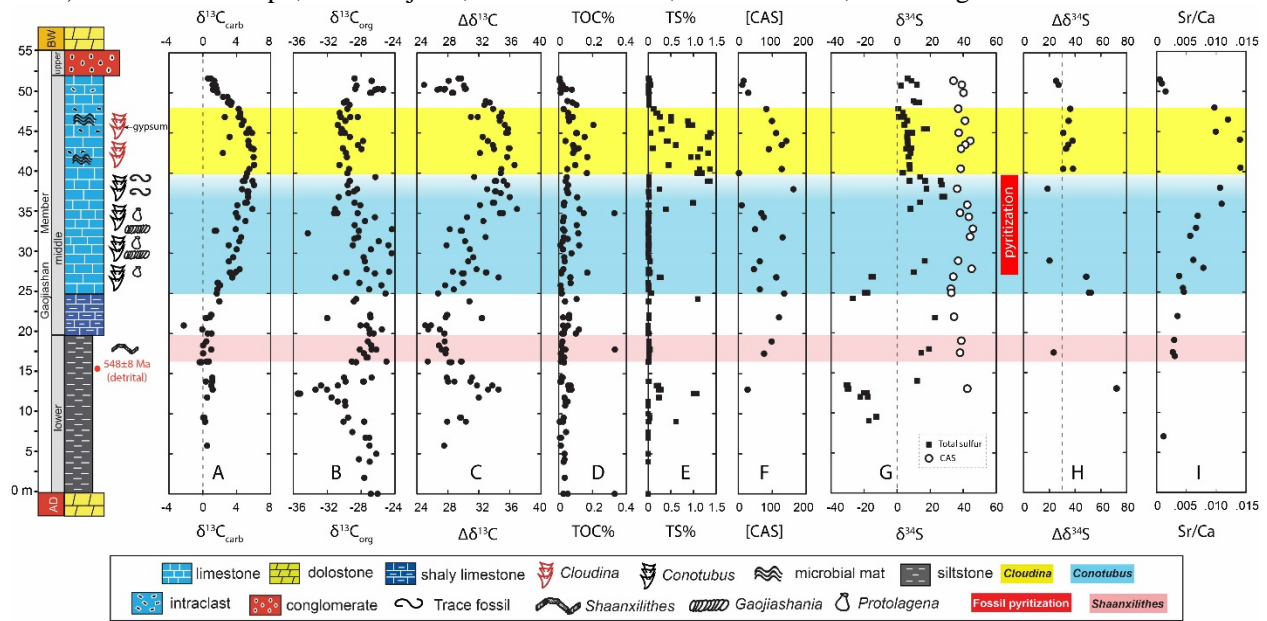


610
 611 **Figure 3:** (A) Field photo showing the position of the two detrital zircon samples (09G-35.3, 09G-37.9) in the
 612 lower Gaojiashan Member. The view is about 3 m wide. (B) Close view of bed 09G-37.9. (C) Separated and
 613 mounted detrital zircon from sample 09G-37.9. (D) Histogram and relative probability plot for detrital zircon
 614 ages from sample 09G-37.9. Best age is $^{206}\text{Pb}/^{238}\text{U}$ date for grains younger than 1 Ga and $^{206}\text{Pb}/^{207}\text{Pb}$ date for
 615 older zircon. Histogram bars represent 50 Ma intervals. Inset: $^{206}\text{Pb}/^{238}\text{U}$ ages for the four youngest analyses
 616 with uncertainties that overlap at the 1-sigma level. The weighted mean of these four ages is 548 ± 8 Ma
 617 (MSWD=0.11) from which we interpret a maximum possible depositional age of 560 Ma.

Environmental context for animal biomineralization

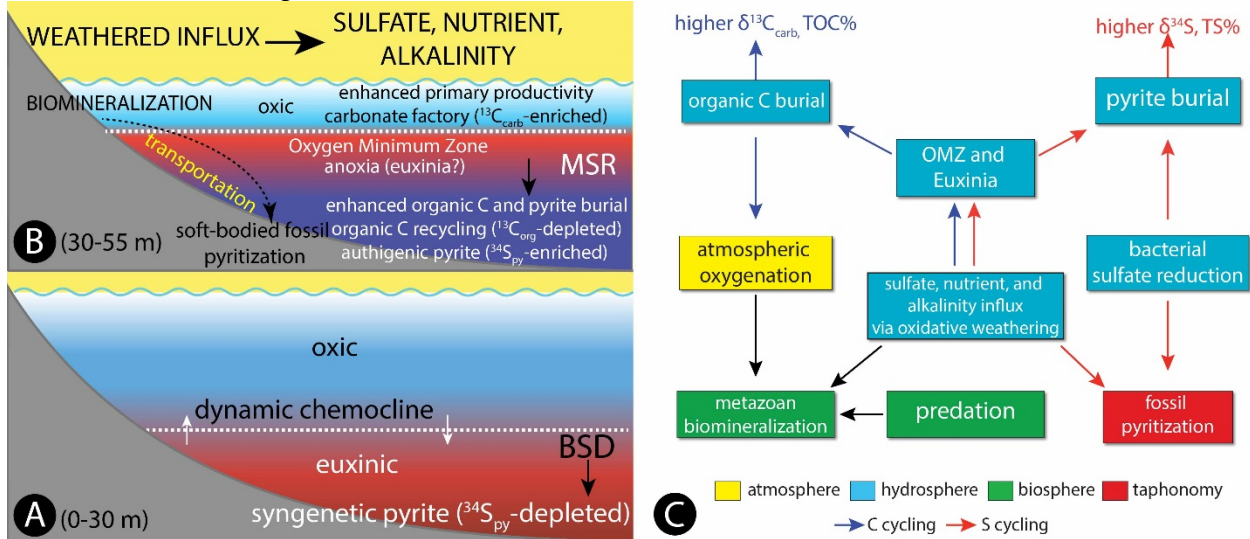


618
 619 **Figure 4:** Integrated litho-, bio-, and chemo-stratigraphy of the Gaojiashan Member, including fossil
 620 occurrences of *Conotubus* and *Cloudina*, as well as geochemical profiles of carbonate content (wt. %),
 621 carbonate carbon ($\delta^{13}\text{C}_{\text{carb}}$, ‰ V-PDB) and oxygen ($\delta^{18}\text{O}_{\text{carb}}$, ‰ V-PDB) isotopes, Sr/Ca, Mg/Ca, Rb/Sr, Mn/Sr,
 622 and Ce anomaly (Ce/Ce^*) calculated using the formula $\text{Ce/Ce}^* = \text{Ce}_{\text{PAAS}} / ([\text{Pr}]^2_{\text{PAAS}} / [\text{Nd}]_{\text{PAAS}})$ (Ling et al.,
 623 2013). K = Kuanchuanpu, G = Guojiaba, Cam = Cambrian, BW = Beiwan, AD = Algal Dolomite.

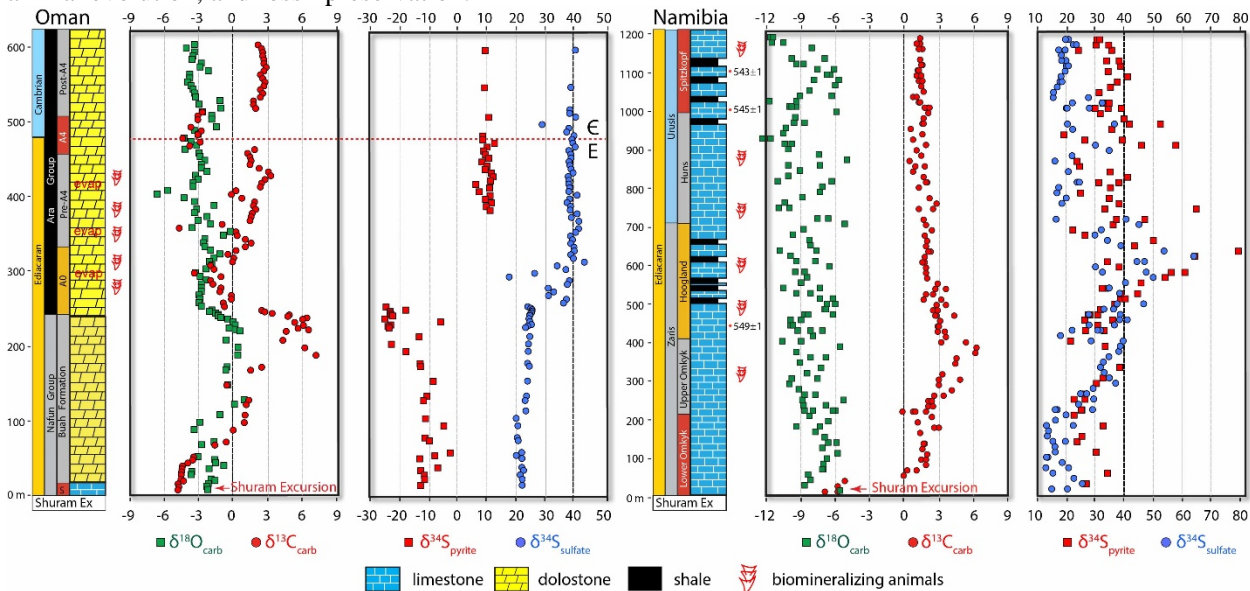


624
 625 **Figure 5:** Integrated litho-, bio-, and chemo-stratigraphy of the Gaojiashan Member, including fossil
 626 occurrences of *Conotubus* and *Cloudina*, as well as geochemical profiles of carbonate carbon ($\delta^{13}\text{C}_{\text{carb}}$, ‰ V-

627 PDB) and organic carbon ($\delta^{13}\text{C}_{\text{org}}$, ‰ V-PDB) isotopes, carbon isotope fractionations ($\Delta\delta^{13}\text{C}_{\text{carb-org}}$), pyrite
 628 sulfur ($\delta^{34}\text{S}_{\text{TS}}$, ‰ V-CDT) and CAS sulfur ($\delta^{34}\text{S}_{\text{CAS}}$, ‰ V-CDT) isotopes, sulfur isotope fractionations
 629 ($\Delta\delta^{34}\text{S}_{\text{CAS-pyrite}}$), total organic carbon content (TOC), total sulfur content (TS, dominated by pyrite with trace
 630 amount of organic S), carbonate-associated sulfate concentration ([CAS] in ppm). The small panels in the
 631 bottom are cross plots of $\delta^{13}\text{C}_{\text{carb}}-\delta^{18}\text{O}_{\text{carb}}$, TOC-TS, $\delta^{13}\text{C}_{\text{org}}-\text{TOC}$, $\delta^{34}\text{S}_{\text{TS}}-\text{TOC}$, $\delta^{34}\text{S}_{\text{TS}}-\text{TS}$, $\delta^{13}\text{C}_{\text{carb}}-\text{Sr}/\text{Ca}$.
 632 BW = Beiwan, AD = Algal Dolomite.

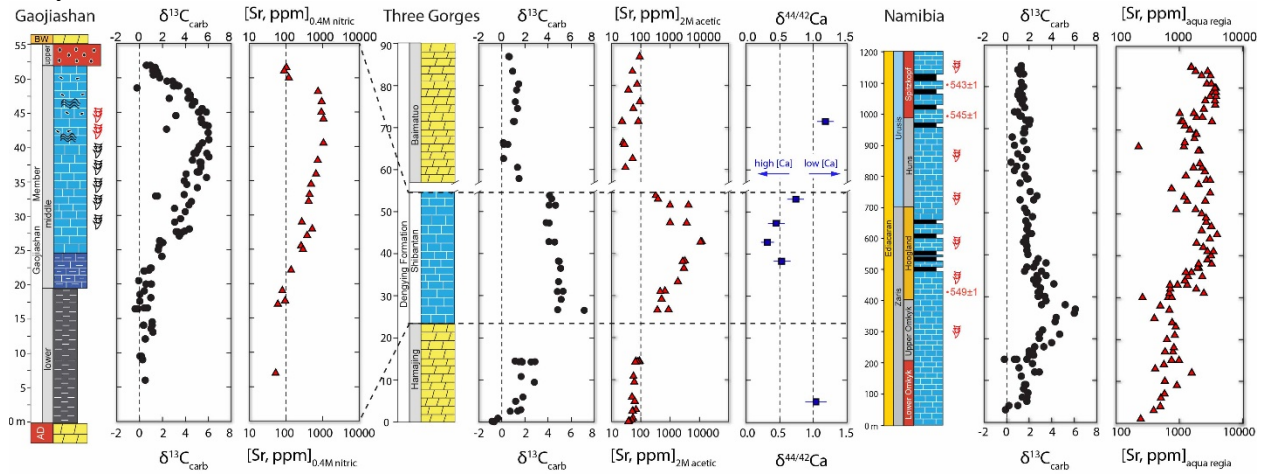


633 **Figure 6.** Conceptual weathering and biogeochemical model for the Gaojiashan Member. (A) During Stage 1
 634 the basin was stratified with oxic surface water above euxinic deep water. Bacterial S disproportionation (BSD)
 635 may be an important contributor to the sulfur cycle. (B) During Stage 2 the basin was strongly influenced by
 636 the spread of an oxygen minimum zone (OMZ) beneath oxic surface waters. This likely occurred as a result of
 637 sea level regression and enhanced continental weathering, which resulted in a larger sulfate pool in the ocean,
 638 elevated ocean alkalinity, and microbial sulfate reduction (MSR) as the dominant pathway for microbial
 639 sulfur cycling. (C) Biogeochemical feedback that link the carbon and sulfur cycles to atmospheric oxygenation,
 640 animal evolution, and fossil preservation.

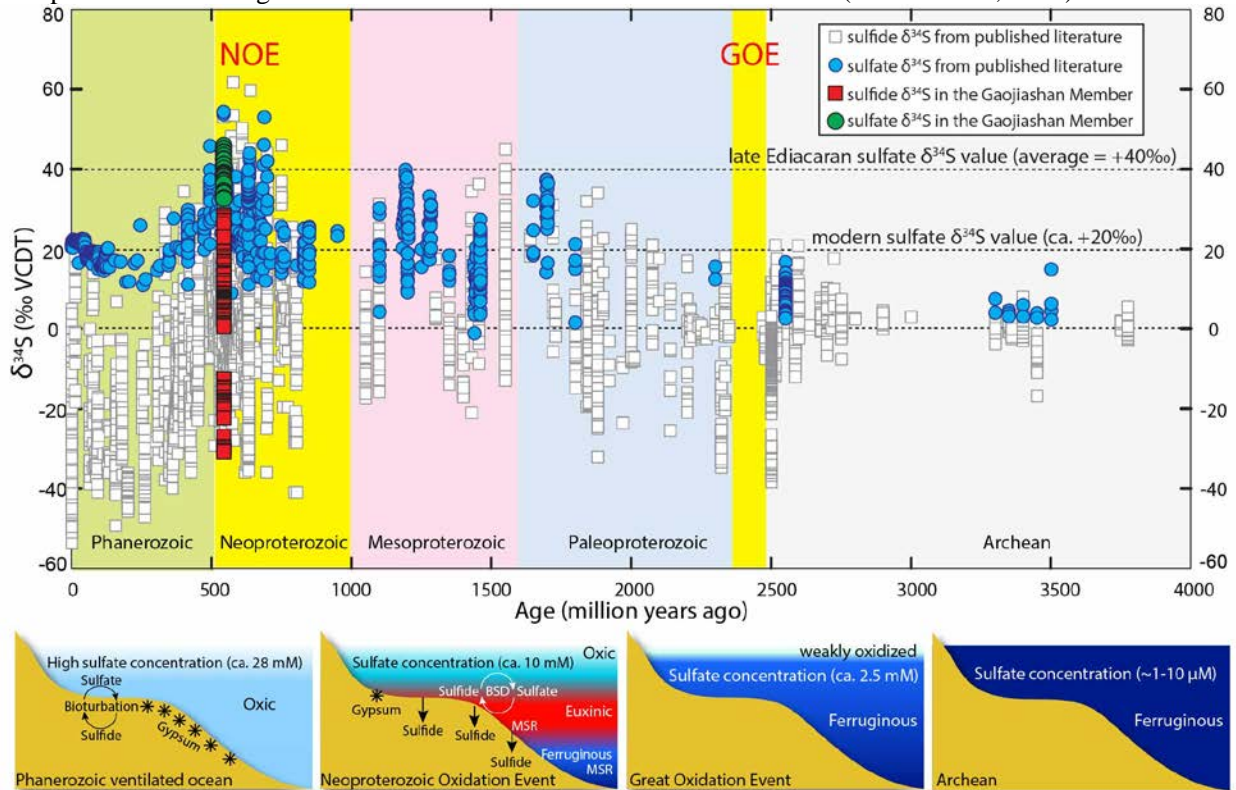


642 **Figure 7:** Integrated carbon and sulfur isotopic profiles from late Ediacaran strata in Oman (Fike and
 643 Grotzinger, 2008) and Namibia (Ries et al., 2009). The remarkable difference in sulfur isotope trends between
 644 Oman and Namibia may result from effects of local conditions (e.g. different organic carbon flux, sulfate
 645

646 concentration) on microbial sulfate reduction rate and sulfur isotope fractionations, but may also reflect
 647 analytical issues (see text).



648
 649 **Figure 8.** Integrated litho-, bio-, and chemo-stratigraphy of the terminal Ediacaran Gaojiashan Member in the
 650 study area, Yangtze Gorges area, and Namibia, including geochemical profiles of carbonate carbon isotopes
 651 ($\delta^{13}\text{C}_{\text{carb}}$, ‰ V-PDB), strontium concentration ([Sr] in ppm, plotted in log scale), and $\delta^{44/42}\text{Ca}$ (‰). Data for the
 652 section in Yangtze Gorges area are from Sawaki et al. (2010) and Sawaki et al. (2014). Data for the section in
 653 Namibia are from Ries et al. (2009). Note that the Sr concentrations are measured by using different acids in
 654 different studies. The negative $\delta^{44}\text{Ca}$ excursion in the Gaojiashan-equivalent Shibantan Member has been
 655 interpreted to reflect high Ca concentration in terminal Ediacaran seawater (Sawaki et al., 2014).



656
 657 **Figure 9.** Evaporite, CAS, and pyrite sulfur isotope data through Earth history. Paired $\delta^{34}\text{S}$ data are compiled
 658 from the literature [after (Canfield and Farquhar, 2009; Och and Shields-Zhou, 2012; Sahoo et al., 2012)].
 659 Recently published Neoproterozoic $\delta^{34}\text{S}$ data have also been included (e.g. Paris et al., 2014; Zhelezinskaia et al.,
 660 2014). The sulfate concentration constraints are ca. 1-10 μM during Archean (Habicht et al., 2002;
 661 Zhelezinskaia et al., 2014), ca. 2.5 mM after the GOE (Shen et al., 2002; Canfield, 2004; Kah et al., 2004;

662 Hurtgen et al., 2005; Bekker et al., 2006; Canfield and Farquhar, 2009; Reuschel et al., 2012), and ca. 10 mM
663 during NOE (Canfield and Farquhar, 2009; Algeo et al., 2015). $\delta^{34}\text{S}_{\text{sulfate}}$ composition of the terminal Ediacaran
664 ocean (ca. +40‰) was determined by measurements of bedded evaporites in Oman (Fike and Grotzinger,
665 2008). GOE = Great Oxidation Event; NOE = Neoproterozoic Oxidation Event. The four conceptual
666 biogeochemical models for redox architectures of the ocean during Archean, GOE, NOE and Phanerozoic are
667 shown in the panels beneath the time-series data. See the main text for detailed discussions.
668

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